

ATOMS



GUIDED READING & PRACTICE WORKSHEETS

ATOMS

1 ELECTRON SHELLS

Q1) An electron shell also known as an energy level, is a region around the nucleus in which electrons are likely to be found. Electrons in an atom are arranged in shells around the nucleus. The number of shells an atom has is equal to the number of protons in the nucleus. The number of electrons in an atom is equal to the number of protons in the nucleus.

Q2) The electrons in an atom fill the lowest energy level first, before moving to the next energy level. This is known as the Aufbau principle. The number of electrons in an atom is equal to the number of protons in the nucleus. The number of shells an atom has is equal to the number of protons in the nucleus.

Q3) Some elements can have different numbers of shells. This is because some elements have a different number of protons than others. The number of shells an atom has is equal to the number of protons in the nucleus.

ATOMS PRACTICE

Complete the following table for each element.

Element	Symbol	Atomic Number	Mass Number	Protons	Neutrons	Electrons
Hydrogen	H	1	1	1	0	1
Helium	He	2	4	2	2	2
Lithium	Li	3	7	3	4	3
Boron	B	5	11	5	6	5
Carbon	C	6	12	6	6	6
Nitrogen	N	7	14	7	7	7
Oxygen	O	8	16	8	8	8
Fluorine	F	9	19	9	10	9
Neon	Ne	10	20	10	10	10

PERIODIC TABLE OF ELEMENTS

Hydrogen (H) is the first element in the periodic table. It has an atomic number of 1 and a mass number of 1. It is a gas at room temperature.

The periodic table is organized into groups and periods. Groups are vertical columns of elements, and periods are horizontal rows of elements. Elements in the same group have similar chemical properties.

The periodic table is divided into three main sections: metals, non-metals, and metalloids. Metals are located on the left side of the table, non-metals on the right side, and metalloids in between.

GRADES
8-12



ATOMS

1 PARTS OF AN ATOM

Q1) Atoms are the basic units of matter and the smallest particles of an element that retain the chemical properties of that element. A particular element is composed of only one type of atom.

Q2) Atoms are made up of three types of subatomic particles: protons, neutrons, and electrons. Protons are positively charged particles that are located in the nucleus of an atom. Neutrons are neutral particles and are also located in the nucleus. Electrons are negatively charged particles that orbit the nucleus in electron shells or energy levels. Typically, the number of protons in an atom is equal to the number of electrons.

Label the atom diagram and fill in the table below.

Diagram of an atom showing the nucleus (protons and neutrons) and electron shells (electrons).

Subatomic Particle	Symbol	Charge	Location
Proton	p^+	Positive	Nucleus
Neutron	n^0	Neutral	Nucleus
Electron	e^-	Negative	Electron Shell

ATOMIC NUMBER & MASS

Q1) The atomic number is the number of protons in the nucleus of an atom. The number of protons in an atom determines what the element is. For example, only atoms with one proton in Hydrogen, while any atom with six protons is Carbon.

Q2) You can find the name of an element on the periodic table by looking at the number of protons in the nucleus. This can also determine the number of protons in an element by looking at the element's atomic number on the periodic table.

Q3) The atomic mass is the sum of the number of protons and neutrons in an atom's nucleus. It is measured in atomic mass units (amu). You can calculate the number of neutrons in an atom by subtracting the number of protons in the nucleus from the atomic mass number. You may round the number to the nearest whole number.

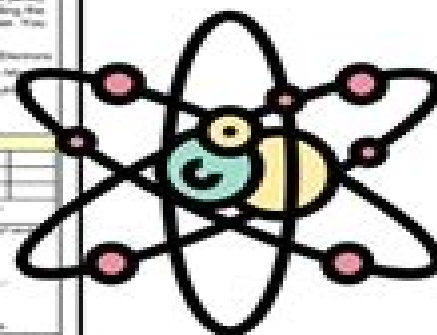
Atomic Number = Number of Protons = Z

Atomic Mass = Number of Protons + Number of Neutrons = A

Number of Neutrons = Atomic Mass - Atomic Number = $A - Z$

Complete the following table for each element.

Element	Symbol	Atomic Number	Mass Number	Protons	Neutrons	Electrons
Hydrogen	H	1	1	1	0	1
Helium	He	2	4	2	2	2
Lithium	Li	3	7	3	4	3
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Carbon	C	6	12	6	6	6
Nitrogen	N	7	14	7	7	7
Oxygen	O	8	16	8	8	8
Fluorine	F	9	19	9	10	9
Neon	Ne	10	20	10	10	10



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Electrons In Atoms Guided Reading 5 2

**Kevin D. Finson, Christine K.
Ormsbee, Mary M. Jensen**



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Atoms, Molecules & Elements Gr. 5-8 George Graybill, 2007-09-01 Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource makes the periodic table easier to understand Begin by answering what are atoms See how the atomic model is made up of electrons protons and neutrons Find out what a molecule is and how they differ from elements Then move on to compounds Find the elements that make up different compounds Get comfortable with the periodic table by recognizing each element as part of a group Examine how patterns in the period table dictate how those elements react with others Finally explore the three important kinds of elements metals nonmetals and inert gases Aligned to the Next Generation Science Standards and written to Bloom s Taxonomy and STEAM initiatives additional hands on experiments crossword word search comprehension quiz and answer key are also included **Atoms, Molecules & Elements: What Are Atoms? Gr. 5-8** George Graybill, 2015-10-01 This is the chapter slice What Are Atoms from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives Atoms, Molecules & Elements: What Are Molecules? Gr. 5-8 George Graybill, 2015-10-01 This is the chapter slice What Are Molecules from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives Atoms, Molecules & Elements: What Are Compounds? Gr. 5-8 George Graybill, 2015-10-01 This is the chapter slice What Are Compounds from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified

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lessons and activities Help is here This timely practical guidebook shows general and special educators how to retool science activities and assessments for students with learning disabilities behavior disorders and more The authors cover a broad range of topics in an orderly concise fashion including National and state requirements for student learning and science literacy Pedagogical strategies for collaborative learning groups self paced learning centers literature circles and team projects Grade appropriate ways to revise science activities and assessments for biology earth science and physical science lessons Step by step instructions for using rubrics for evaluation revision and assessment Information on teacher collaboration and specific disabilities Also included are vignettes and checklists to assist teachers in bridging the gap between science and special education instruction and assessment By adjusting the content teaching critical thinking and providing a variety of ways for learners to demonstrate their knowledge you will give all students the chance to achieve academic success in science

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Structure of Multielectron Atoms Anders Kastberg, 2020-04-23 This book is about the structure of multielectron atoms and predominantly adopts a perturbative approach to the total Hamiltonian A key concept is the central field approximation and beyond the standard LS coupling and jj coupling schemes intermediate cases are also treated After that the book covers hyperfine structure and other nuclear effects as well as interactions with static external fields Throughout the book an analytical approach is adopted Working knowledge of basic quantum mechanics including the non relativistic hydrogen atom basic angular momentum and perturbation theory is assumed and it begins with a brief recap of the hydrogen orbitals before turning towards the symmetry aspects of multi electron atoms spin orbit interaction and couplings of angular momenta

Readers' Guide to Periodical Literature, 1919

A Student's Guide Through the Great Physics Texts Kerry Kuehn, 2015-12-17 This book provides a chronological introduction to modern atomic theory which represented an attempt to reconcile the ancient doctrine of atomism with careful experiments performed during the 19th century on the flow of heat through substances and across empty space Included herein are selections from classic texts such as Carnot's Reflection on the Motive Power of Fire Clausius Mechanical Theory of Heat Rutherford's Nuclear Constitution of Atoms Planck's Atomic Theory of Matter and Heisenberg's Copenhagen Interpretation of Quantum Theory Each chapter begins with a short introduction followed by a reading selection Carefully crafted study questions draw out key points in the text and focus the reader's attention on the author's methods analysis and conclusions Numerical and laboratory exercises at the end of each chapter test the reader's ability to understand and apply key concepts from the text Heat Radiation and Quanta is the last of four volumes in A Student's Guide through the Great Physics Texts The book comes from a four semester undergraduate physics curriculum designed to encourage a critical and circumspect approach to natural science while at the

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Technical Abstract Bulletin Defense Documentation Center (U.S.),1963 *Cambridge Checkpoint Lower Secondary Science Revision Guide for the Secondary 1 Test 2nd edition* Rosemary Feasey,Andrea Mapplebeck,David Bailey,2023-04-06 Achieve maximum potential using step by step guidance that helps to practise skills learned and improve exam technique Build confidence with practical study tips and effective revision strategies Reinforce understanding with clear explanations of every topic covered in the Cambridge Lower Secondary Checkpoint curriculum frameworks Strengthen and test knowledge with a range of questions and worked examples Test questions worked solutions and answers are available free online at www.hoddereducation.co.uk/cambridgeextras This resource has not been through the Cambridge International endorsement process *HP TGT Non - Medical Exam Book (English Edition) | Himachal Pradesh - Trained Graduate Teacher | 10 Practice Tests (1500 Solved MCQ)* , Best Selling Book in English Edition for Himachal Pradesh HP TGT Non Medical Exam with objective type questions as per the latest syllabus Himachal Pradesh HP TGT Non Medical Exam Preparation Kit comes with 10 Practice Tests with the best quality content Increase your chances of selection by 16X Himachal Pradesh HP TGT Non Medical Exam Prep Kit comes with well structured and 100% detailed solutions for all the questions Clear exam with good grades using thoroughly Researched Content by experts

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