

5

ELECTRONS IN ATOMS

Practice Problems

In your notebook, solve the following problems.

SECTION 5.1 MODELS OF THE ATOM

- How many sublevels are in the following principal energy levels?

a. $n = 1$	c. $n = 3$	e. $n = 5$
b. $n = 2$	d. $n = 4$	f. $n = 6$
- How many orbitals are in the following sublevels?

a. 1s sublevel	d. 4f sublevel	g. fifth principal energy level
b. 5s sublevel	e. 7s sublevel	h. 6d sublevel
c. 4d sublevel	f. 3p sublevel	
- What are the types of sublevels and number of orbitals in the following energy levels?

a. $n = 1$	c. $n = 3$	e. $n = 5$
b. $n = 2$	d. $n = 4$	

SECTION 5.2 ELECTRON ARRANGEMENT IN ATOMS

- Write a complete electron configuration of each atom.

a. hydrogen	d. barium	g. krypton
b. vanadium	e. bromine	h. arsenic
c. magnesium	f. sulfur	i. radon

SECTION 5.3 PHYSICS AND THE QUANTUM MECHANICAL MODEL

- What is the wavelength of the radiation whose frequency is $5.00 \times 10^{15} \text{ s}^{-1}$? In what region of the electromagnetic spectrum is this radiation?
- An inexpensive laser that is available to the public emits light that has a wavelength of 670 nm. What are the color and frequency of the radiation?
- What is the energy of a photon whose frequency is $2.22 \times 10^{14} \text{ s}^{-1}$?
- What is the frequency of a photon whose energy is $6.00 \times 10^{-19} \text{ J}$?
- Arrange the following types of electromagnetic radiation in order of increasing frequency.

a. infrared	c. visible light	e. microwaves
b. cosmic rays	d. radio waves	f. ultraviolet
- Suppose that your favorite AM radio station broadcasts at a frequency of 1500 kHz. What is the wavelength in meters of the radiation from the station?

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J. S. Wright



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the Next Generation Science Standards and written to Bloom s Taxonomy and STEAM initiatives additional hands on experiments crossword word search comprehension quiz and answer key are also included

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Behavior of Electrons in Atoms Robin M. Hochstrasser,1964 Atomic spectra The interactions between atoms and electrons Quantum theory of atomic structure The Pauli principle and the electronic structure of atoms Energy terms and states of atoms Atomic excitation probabilities Collisional processes involving excited atoms The behavior of atoms in magnetic fields Some of the forces between atoms The simplest molecules The Basics of Atoms and Molecules Tom Jackson,2013-12-15 A fun filled introduction to matter the elements of the periodic table atoms electrons reactions and bonding and radioactivity this volume provides young adults with chemistry examples that reflect their real world interconnections in science Key terms easy experiments and clear illustrations help to guide students through chemical applications A chapter about Niels Bohr and his model for the atom honors his contribution to the understanding of atomic structure and to nuclear fission Tools and techniques such as a scanning tunneling microscope Rutherford s gold foil experiment and a mass spectrometer highlight this instructive text that is aligned to the Common Core Standards

Electrons in Atoms Karen Horney,1963 **Electrons in Atoms** G. F. Lothian,1963 **Understanding Electrons** B. H. Fields,Fred Bortz,2015-07-15 Explore the history of the discovery and the properties of the electron the negatively charged subatomic particle that orbits around an atom s nucleus

Electrons and Atoms John L. Lewis,1972 *Atoms and Electrons* John William Navin Sullivan,1895* *The Number of Electrons in Atoms* Harold Albert Wilson,1912

Atoms, Electrons, Structure and Bonding Primrose Kitten,2017-06-19 I m constantly telling you the best way to learn is by practicing questions so I ve made you a book full of practice questions Multiple choice questions to reflect the style of exam questions activities to complete equations for you to balance compounds for you to work out the formula for lots of things that you need to recall and practice long answer exam style questions This book is not designed as a text book or revision guide but as a workbook There are lots of good and bad expensive and free revision guides out there on my

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