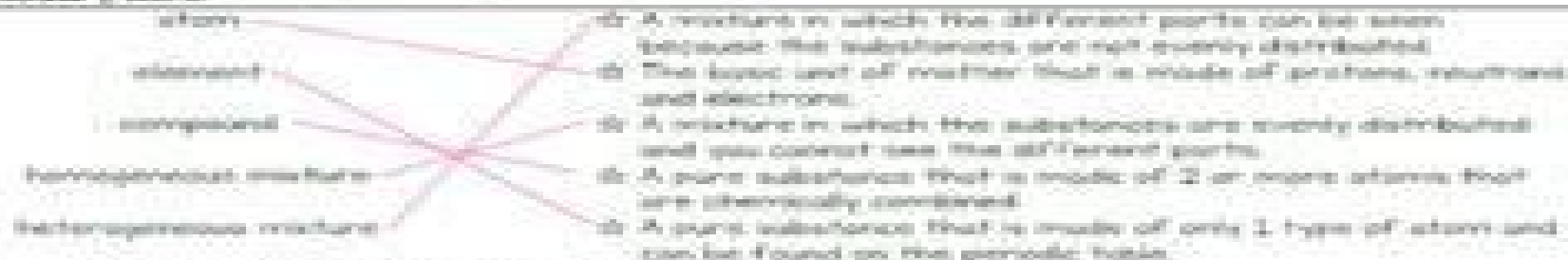


PROPERTIES OF MATTER STUDY GUIDE



ATOMS, ELEMENTS, COMPOUNDS + MIXTURES

Complete the vocabulary matching below by drawing a line to match the correct definition to the correct vocabulary word.



Complete the chart below about the subatomic particles and their charges:

	PARTICLE #1	PARTICLE #2	PARTICLE #3
NAME	proton	neutron	electron
CHARGE	+ (positive)	0 (neutral)	- (negative)
LOCATION	part of nucleus	part of nucleus	outer "cloud-like" orbit

Complete the graphic organizer below about matter below. Need help? Use the word bank to assist you!

heterogeneous	element	mixture	compound
	pure substance	electron	homogeneous



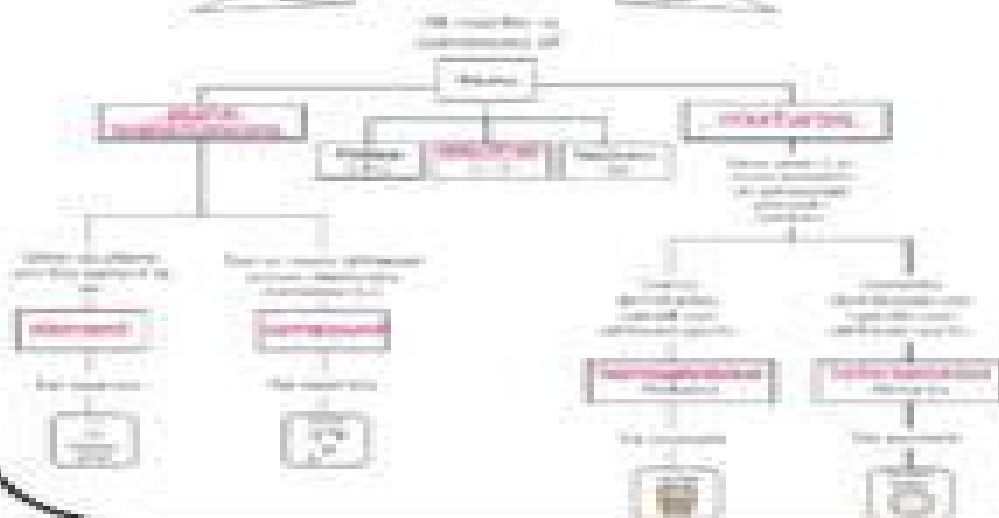
Classify the following examples as elements, compounds or mixtures in the table below.

carbon (C)	salt	water (H ₂ O)
hydrogen (H)	salt water	sugar (sucrose)

ELEMENT	COMPOUND	MIXTURE
carbon (C) hydrogen (H)	water (H ₂ O) salt (NaCl) sugar (sucrose)	salt

Why is carbon considered an element and carbon dioxide considered a compound?

Carbon is an element on the periodic table, made of only carbon atoms. Carbon dioxide is carbon and oxygen chemically combined together.



Matter Review Guide Answer Key

**Susan K. Grove, Jennifer R. Gray, PhD,
RN, FAAN, Nancy Burns**



Matter Review Guide Answer Key:

Official Guide to Mastering the DSST Peterson's, 2010-08-01 Peterson's Official Guide to Mastering the DSST Exams helps nontraditional students earn college credits for life and learning experiences with diagnostic tests subject review and post tests with detailed answer explanations for each of the 8 most popular DSST exams Ethics in America Introduction to Computing Principles of Supervision Substance Abuse Business Math Principles of Public Speaking Fundamentals of College Algebra and Technical Writing Peterson's Official Guide to Mastering the DSST Exams is the only prep guide endorsed by Prometric the DSST program provider which found this study guide to be an excellent reflection of the content of the respective DSST tests

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