3.1 - The Periodic Table

3.1.1 - Describe the arrangement of elements in the periodic table in order of increasing atomic number

Elements in the periodic table are arranged in order of **increasing atomic number** (Z). There is a division between metals and non-metals. Metals are on the left and non-metals are on the right. Metals tend to have a smaller number of electrons in their outer shell

The long metal periods are divided into the transition metals, lanthanides and actinides

Hydrogen is difficult to place, however it is place in group 1 because it displays some of the same characteristics as these elements, although it is a non-metal

Helium is still a noble gas because its outer shell is filled by 2 electrons

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