

Stoichiometry Study Guide

Students should be able to...

- Explain mole ratio
- Convert moles, grams, atoms/molecules
- Calculate theoretical yields when given the limiting reactant (Convert: moles A to moles B, grams A to moles B, moles A to grams B, grams A to grams B)
- Explain limiting reactant and excess.
- Calculate theoretical yields by finding the limiting reactant
- Calculate percent yield.
- Calculate percent composition.
- Calculate theoretical yield, limiting reactant, excess, and percent yield for a double replacement reaction (laboratory)

Practice Problems

Find the molar mass of each compound.

1. BaI_2
2. $\text{AgC}_2\text{H}_3\text{O}_2$

Convert.

3. 0.4 moles of Sn to atoms
4. 32 g of K_2O to molecules
5. 3.7 g of H_2 to moles
6. 800 g of CaCl_2 to moles
7. 0.75 moles of H_2O to grams
8. 9.0 g of Cl_2 to atoms
9. 0.5 moles of KNO_3 to grams

Find the percent composition for each element in the compound.

10. Mn_2O_7
11. AuNO_3

Stoichiometry



If you started with 3.7 moles of magnesium oxide, how many moles of magnesium hydroxide would be produced?



If 2.8 grams of sodium were used for the reaction, and an excess of iodine was present, how many moles of sodium iodide would be produced?



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M Tight



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web the robot is featured in the nxt 2 0 inventor s guide building and programming this robot instructions for building and

programming this robot are included in the unofficial lego mindstorms nxt 2.0 inventor's guide english and in lego roboter ii german

[nxt bumper car nxt programs](#) - Mar 13 2023

web building instructions start by building the castor bot click the picture for building instructions use the shortest wire for this step connect the wire from the touch sensor to port 1 on the nxt you can route the wire around the round cross brace under the robot to keep it out of the way use the program bumper car for your bumper car