

Wintersemester 2019/2020 - Klausuraufgaben 8B

Aufgabe 8 (max. 70 Punkte/100% Gesamtpunkte)

Die nachstehenden Probleme betreffen die Veranschaulichung des Verhaltens der Stoffeigenschaften mittels "Diagramme" von Molanteile, Druck, Temperatur in der Zustandsform. Bitte führen Sie alle Schritte der Berechnungen und die Skizzen an!

$$M(\text{H}_2\text{O}) = 18,015 \text{ g/mol}$$

1. Wasser wird bei 100°C und $1013,25 \text{ hPa}$ verdampft. Wie groß ist die spezifische Verdampfungswärme $q_{\text{v,100}}^{\text{Wasser}}$?

$$q_{\text{v,100}}^{\text{Wasser}} = \frac{h_{\text{v,100}}^{\text{Wasser}}}{M(\text{H}_2\text{O})} = 2257 \text{ kJ/kg}$$

2. Wasserdampf wird bei 100°C und $1013,25 \text{ hPa}$ verdampft. Wie groß ist die spezifische Verdampfungswärme $q_{\text{v,100}}^{\text{Wasser}}$?

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3. Wasser wird bei 100°C und $1013,25 \text{ hPa}$ verdampft. Wie groß ist die spezifische Verdampfungswärme $q_{\text{v,100}}^{\text{Wasser}}$?

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4. In der Zustandsform des flüssigen Wassers bei 100°C und $1013,25 \text{ hPa}$ wird die spezifische Verdampfungswärme $q_{\text{v,100}}^{\text{Wasser}}$ bestimmt. Wie groß ist die spezifische Verdampfungswärme $q_{\text{v,100}}^{\text{Wasser}}$?

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6. In der Zustandsform des flüssigen Wassers bei 100°C und $1013,25 \text{ hPa}$ wird die spezifische Verdampfungswärme $q_{\text{v,100}}^{\text{Wasser}}$ bestimmt. Wie groß ist die spezifische Verdampfungswärme $q_{\text{v,100}}^{\text{Wasser}}$?

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