

Name: _____

Date: _____

Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? **They can share the markers.**

2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

They can give the ones they need to each other.

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have? _____7_____
2. Click **Pause** (). Drag an electron from the left atom to the right atom. Click **Play**.
What happens? **They share an electron.**
3. Click **Pause**, drag an electron from the right atom to the left, and then click **Play**.
What happens now? **They share another one.**

Gizmo Answer Key Covalent Bonds

Victor M. Corman

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Gizmo Answer Key Covalent Bonds

1. Understanding the eBook Gizmo Answer Key Covalent Bonds
 - The Rise of Digital Reading Gizmo Answer Key Covalent Bonds
 - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Answer Key Covalent Bonds
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gizmo Answer Key Covalent Bonds
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmo Answer Key Covalent Bonds
 - Personalized Recommendations

- Gizmo Answer Key Covalent Bonds User Reviews and Ratings
- Gizmo Answer Key Covalent Bonds and Bestseller Lists
- 5. Accessing Gizmo Answer Key Covalent Bonds Free and Paid eBooks
 - Gizmo Answer Key Covalent Bonds Public Domain eBooks
 - Gizmo Answer Key Covalent Bonds eBook Subscription Services
 - Gizmo Answer Key Covalent Bonds Budget-Friendly Options
- 6. Navigating Gizmo Answer Key Covalent Bonds eBook Formats
 - ePub, PDF, MOBI, and More
 - Gizmo Answer Key Covalent Bonds Compatibility with Devices
 - Gizmo Answer Key Covalent Bonds Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmo Answer Key Covalent Bonds
 - Highlighting and Note-Taking Gizmo Answer Key Covalent Bonds
 - Interactive Elements Gizmo Answer Key Covalent Bonds
- 8. Staying Engaged with Gizmo Answer Key Covalent Bonds
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gizmo Answer Key Covalent Bonds
- 9. Balancing eBooks and Physical Books Gizmo Answer Key Covalent Bonds
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmo Answer Key Covalent Bonds
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gizmo Answer Key Covalent Bonds
 - Setting Reading Goals Gizmo Answer Key Covalent Bonds
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gizmo Answer Key Covalent Bonds
 - Fact-Checking eBook Content of Gizmo Answer Key Covalent Bonds

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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