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Student Exploration: Ionic Bonds

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

If Nate were to give Clara one of his markers, then both would end up with eight.

2. Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

If Maggie gives Fred and Florence one marker each, they will all have eight markers.

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The *Ionic Bonds* Gizmo allows you to explore how **ionic bonds** form.

To begin, check that **Sodium (Na)** and **Chlorine (Cl)** are selected from the menus at right. Click **Play** () to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



1. Each atom consists of a central nucleus and several **shells** that contain **electrons**. The outermost electrons are called **valence electrons**.

How many valence electrons does each atom have? Sodium: 1 Chlorine:

7

2. Click **Pause** (). Elements can be classified as **metals** and **nonmetals**. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. **Electron affinity** is a measure of how tightly the valence electrons are held.

A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

Gizmo Warmup Ionic Bonds Answers

J Elliott

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Gizmo Warmup Ionic Bonds Answers

1. Understanding the eBook Gizmo Warmup Ionic Bonds Answers
 - The Rise of Digital Reading Gizmo Warmup Ionic Bonds Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Warmup Ionic Bonds Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gizmo Warmup Ionic Bonds Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmo Warmup Ionic Bonds Answers
 - Personalized Recommendations
 - Gizmo Warmup Ionic Bonds Answers User Reviews and Ratings
 - Gizmo Warmup Ionic Bonds Answers and Bestseller Lists
5. Accessing Gizmo Warmup Ionic Bonds Answers Free and Paid eBooks
 - Gizmo Warmup Ionic Bonds Answers Public Domain eBooks
 - Gizmo Warmup Ionic Bonds Answers eBook Subscription Services

- Gizmo Warmup Ionic Bonds Answers Budget-Friendly Options
- 6. Navigating Gizmo Warmup Ionic Bonds Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Gizmo Warmup Ionic Bonds Answers Compatibility with Devices
 - Gizmo Warmup Ionic Bonds Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmo Warmup Ionic Bonds Answers
 - Highlighting and Note-Taking Gizmo Warmup Ionic Bonds Answers
 - Interactive Elements Gizmo Warmup Ionic Bonds Answers
- 8. Staying Engaged with Gizmo Warmup Ionic Bonds Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gizmo Warmup Ionic Bonds Answers
- 9. Balancing eBooks and Physical Books Gizmo Warmup Ionic Bonds Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmo Warmup Ionic Bonds Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gizmo Warmup Ionic Bonds Answers
 - Setting Reading Goals Gizmo Warmup Ionic Bonds Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gizmo Warmup Ionic Bonds Answers
 - Fact-Checking eBook Content of Gizmo Warmup Ionic Bonds Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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