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Student Exploration: Ionic Bonds

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

If Nate were to give Clara one of his markers, then both would end up with eight.

2. Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

If Maggie gives Fred and Florence one marker each, they will all have eight markers.

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The *Ionic Bonds* Gizmo allows you to explore how **ionic bonds** form.

To begin, check that **Sodium (Na)** and **Chlorine (Cl)** are selected from the menus at right. Click **Play** () to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



1. Each atom consists of a central nucleus and several **shells** that contain **electrons**. The outermost electrons are called **valence electrons**.

How many valence electrons does each atom have? Sodium: 1 Chlorine:

7

2. Click **Pause** (). Elements can be classified as **metals** and **nonmetals**. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. **Electron affinity** is a measure of how tightly the valence electrons are held.

A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

Gizmo Student Exploration Covalent Bonds

Julia Schneider



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Table of Contents Gizmo Student Exploration Covalent Bonds

1. Understanding the eBook Gizmo Student Exploration Covalent Bonds
 - The Rise of Digital Reading Gizmo Student Exploration Covalent Bonds
 - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Student Exploration Covalent Bonds
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gizmo Student Exploration Covalent Bonds
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmo Student Exploration Covalent Bonds
 - Personalized Recommendations
 - Gizmo Student Exploration Covalent Bonds User Reviews and Ratings
 - Gizmo Student Exploration Covalent Bonds and Bestseller Lists

5. Accessing Gizmo Student Exploration Covalent Bonds Free and Paid eBooks
 - Gizmo Student Exploration Covalent Bonds Public Domain eBooks
 - Gizmo Student Exploration Covalent Bonds eBook Subscription Services
 - Gizmo Student Exploration Covalent Bonds Budget-Friendly Options
6. Navigating Gizmo Student Exploration Covalent Bonds eBook Formats
 - ePub, PDF, MOBI, and More
 - Gizmo Student Exploration Covalent Bonds Compatibility with Devices
 - Gizmo Student Exploration Covalent Bonds Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmo Student Exploration Covalent Bonds
 - Highlighting and Note-Taking Gizmo Student Exploration Covalent Bonds
 - Interactive Elements Gizmo Student Exploration Covalent Bonds
8. Staying Engaged with Gizmo Student Exploration Covalent Bonds
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gizmo Student Exploration Covalent Bonds
9. Balancing eBooks and Physical Books Gizmo Student Exploration Covalent Bonds
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmo Student Exploration Covalent Bonds
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Gizmo Student Exploration Covalent Bonds
 - Setting Reading Goals Gizmo Student Exploration Covalent Bonds
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Gizmo Student Exploration Covalent Bonds
 - Fact-Checking eBook Content of Gizmo Student Exploration Covalent Bonds
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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