

Pairs of electrons	BP	LP	Shape	Bond angle	E.g.
2	2	0	Linear	180°	CO <sub>2</sub>
3	3	0	<u>Trigonal planar</u>	120°	BF <sub>3</sub>
4	4	0	Tetrahedral	109.5°	CH <sub>4</sub>
4	3	1	Pyramidal	107°	NH <sub>3</sub>
4	2	2	Non-linear	104.5°	H <sub>2</sub> O
6	6	0	Octahedral	90°	SF <sub>6</sub>

F321 Ocr 2014

**Puja Mehta**



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