

| Question | | | Answer | Mark | Guidance |
|----------|-----|------|---|------|--|
| 2 | (a) | | FIRST CHECK THE ANSWER ON THE ANSWER LINE IF answer = $\text{CH}_4\text{N}_2\text{O}$ award 2 marks C H N O 20.00/12.0 6.67/1.0 46.67/14.0 26.66/16.0 OR 1.67 6.67 3.33 1.67 ratio of mol ✓ to give $\text{CH}_4\text{N}_2\text{O}$ ✓ | 2 | ALLOW 1.66 for C OR 1.66 for O IGNORE Significant figures beyond the 3rd significant figure, (eg ALLOW 3.3335 for N OR 1.666 for C) ALLOW ECF from incorrectly calculated ratio of mol, DO NOT ALLOW ECF from using an atomic number OR any original sums inverted (eg 12.00/20.00) ALLOW any order of atoms |
| 2 | (b) | | NH_4^+ ✓ NO_3^- ✓ | 2 | Mark incorrect ions first |
| 2 | (c) | (i) | H_3PO_4 ✓ | 1 | ALLOW formula if seen as reactant in an equation IGNORE name |
| 2 | (c) | (ii) | Calcium oxide OR calcium hydroxide OR calcium carbonate ✓ | 1 | IGNORE formulae IGNORE lime, quicklime and limestone |
| Total | | | | 6 | |

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