

# 2.5

## ELEMENTS AND COMPOUNDS

### Section Review

#### Objectives

1. Explain the difference between elements and a compound
2. Distinguish between a substance and a mixture
3. Identify the chemical symbols and formulas and write a chemical formula for a substance

#### Vocabulary

- |              |                |
|--------------|----------------|
| 1. element   | 2. compound    |
| 3. substance | 4. homogeneous |

#### Part A Completion

Fill in the blank with the word or phrase that will complete the statement correctly. Write your answer in the space provided. Write the chemical symbol or formula for the element or compound in the space provided.

- |   |   |
|---|---|
| 1. An element is a substance that cannot be broken down into simpler substances by chemical means. _____                | 2. A compound is a substance that is made up of two or more different elements chemically combined. _____                       |
| 3. A mixture is a combination of two or more substances that are not chemically combined. _____                         | 4. A homogeneous mixture is a mixture in which the components are uniformly distributed. _____                                  |
| 5. A heterogeneous mixture is a mixture in which the components are not uniformly distributed. _____                    | 6. A solution is a homogeneous mixture of two or more substances. _____   |
| 7. A pure substance is a substance that has a constant composition and distinct chemical and physical properties. _____ | 8. A mixture is a combination of two or more substances that are not chemically combined. _____                                 |
| 9. A chemical formula is a symbolic representation of a chemical compound. _____  | 10. A chemical symbol is a one- or two-letter abbreviation for an element. _____  |
| 11. A chemical equation is a symbolic representation of a chemical reaction. _____                                      | 12. A chemical reaction is a process in which one or more substances are converted into one or more different substances. _____ |
| 13. A chemical formula is a symbolic representation of a chemical compound. _____                                       | 14. A chemical symbol is a one- or two-letter abbreviation for an element. _____  |
| 15. A chemical equation is a symbolic representation of a chemical reaction. _____                                      | 16. A chemical reaction is a process in which one or more substances are converted into one or more different substances. _____ |

# Elements And Compounds Section Review

**R Sanford**



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