

Activity A:

Subatomic particles

Get the Gizmo ready:

- Use the arrows to create an atom with two protons, two neutrons, and two electrons.
- Turn on **Show element name**.



Question: What are the properties of protons, neutrons, and electrons?

1. **Observe:** Turn on **Show element symbol** and **Element notation**. Three numbers surround the element symbol: the **mass number** (A), electrical charge (no number is displayed if the atom is neutral), and the **atomic number** (Z).
2. **Investigate:** Watch how the numbers change as you add or remove particles.



- A. Which number is equal to the number of protons in the atom?
- B. How can you calculate the number of neutrons (N) in an atom?
- C. Which particle (proton, neutron, or electron) has a
Positive charge?
Negative charge?

neutrons

by the number of protons

electron

proton

Electrical charge?

3. **Observe:** Turn on **Electron configuration**. An atom's electrons form an electron cloud. Electrons are arranged in shells. The number of shells is equal to the atomic number. The number of electrons in each shell is also equal to the atomic number. The number of electrons in each shell is also equal to the atomic number. The number of electrons in each shell is also equal to the atomic number.

- A. What are the shells called?
- B. What are the shells called?
- C. What are the shells called?

Electron shells

Electron shells

Electron shells

4. **Observe:** Turn on **Electron configuration**. The following questions are asked:

- A. How many electrons are in the outer shell of Helium?
- B. How many electrons are in the outer shell of Hydrogen?
- C. What is the mass number of an atom with 2 protons and 2 neutrons?
- D. What is the mass number of an atom with 2 protons and 2 neutrons?
- E. How many electrons are in Helium?

2

1

4

4

2

Element Builder Gizmo Answers Key

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Table of Contents Element Builder Gizmo Answers Key

1. Understanding the eBook Element Builder Gizmo Answers Key
 - The Rise of Digital Reading Element Builder Gizmo Answers Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Element Builder Gizmo Answers Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Element Builder Gizmo Answers Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Element Builder Gizmo Answers Key
 - Personalized Recommendations
 - Element Builder Gizmo Answers Key User Reviews and Ratings
 - Element Builder Gizmo Answers Key and Bestseller Lists
5. Accessing Element Builder Gizmo Answers Key Free and Paid eBooks
 - Element Builder Gizmo Answers Key Public Domain eBooks
 - Element Builder Gizmo Answers Key eBook Subscription Services

- Element Builder Gizmo Answers Key Budget-Friendly Options
- 6. Navigating Element Builder Gizmo Answers Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Element Builder Gizmo Answers Key Compatibility with Devices
 - Element Builder Gizmo Answers Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Element Builder Gizmo Answers Key
 - Highlighting and Note-Taking Element Builder Gizmo Answers Key
 - Interactive Elements Element Builder Gizmo Answers Key
- 8. Staying Engaged with Element Builder Gizmo Answers Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Element Builder Gizmo Answers Key
- 9. Balancing eBooks and Physical Books Element Builder Gizmo Answers Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Element Builder Gizmo Answers Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Element Builder Gizmo Answers Key
 - Setting Reading Goals Element Builder Gizmo Answers Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Element Builder Gizmo Answers Key
 - Fact-Checking eBook Content of Element Builder Gizmo Answers Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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