

Chapter 1

Introduction to Electronics

Question A.1: Atomic Structure

- (a) The atom is the smallest constituent of matter that retains all the chemical properties of the element.
- (b) The atom consists of three subatomic particles: electrons, protons, and neutrons.

Question A.2: Materials Used for Electronics

- (a) The materials are generally classified into three categories: (i) conductors, (ii) semiconductors, and (iii) insulators.
- (b) The most commonly used semiconductor is silicon.
- (c) An atom is made up of three subatomic particles: electrons, protons, and neutrons.

Question A.3: Current in Semiconductors

- (a) Current is the flow of electric charge. In semiconductors, it is carried by both electrons and holes.
- (b) Current is produced when an electric field is applied across the semiconductor.

Question A.4: N-Type and P-Type Semiconductors

- (a) In an N-type semiconductor, the majority carriers are electrons, which are produced by the addition of pentavalent impurities to the intrinsic semiconductor. The impurities have five valence electrons, one more than the silicon atoms.
- (b) In a P-type semiconductor, the majority carriers are holes, which are produced by the addition of trivalent impurities to the intrinsic semiconductor. The impurities have three valence electrons, one less than the silicon atoms.

Question A.5: The PN Junction

- (a) The PN junction is formed by joining an N-type semiconductor with a P-type semiconductor. The junction is formed by the diffusion of dopant atoms from the N-type region into the P-type region, and vice versa. The junction is formed by the diffusion of dopant atoms from the P-type region into the N-type region, and vice versa.
- (b) The PN junction is a semiconductor device that is used to convert electrical energy into light energy. It is used in LEDs, laser diodes, and photodiodes.

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Shasha Hu



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