

Name : _____

Electron Configuration Worksheet

1. Name the elements that have the following electron configurations.

i. $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^5$ _____

ii. $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^4$ _____

2. Write the ground state electron configuration for a neutral atom of the following elements.

i. Oxygen _____

ii. Krypton _____

iii. Chromium _____

3. Using the long method, give the electron configuration of

i. Magnesium (Mg) _____

ii. Potassium (K) _____

iii. Lithium (Li) _____

iv. Nickel (Ni) _____

v. Sulfur (S) _____

4. Identify the following elements.

i. $1s^2 2s^2 2p^2$ _____

ii. $1s^2 2s^2 2p^6$ _____

iii. $[\text{Ar}] 4s^2 3d^{10} 4p^4$ _____

iv. $[\text{Kr}] 5s^2 4d^1$ _____

v. $[\text{Ne}] 3s^2 3p^3$ _____

5. Determine the electron configuration using the short method.

i. Strontium _____

ii. Bromine _____

iii. Zirconium _____

iv. Molybdenum _____

v. Silver _____

Electronic Configuration Worksheet 75 Questions And Answers

N Colangelo



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Exploration of the Structure of Atom Chandan Sengupta, First Publication April 2019 Revised Publication October 2022
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