Practice Problems (Chapter 8): Electron Configuration

CHEM IA

(3d el are core el

for rep. elements)

1. Write complete electron configurations for the following atoms:

a) F 1s² 2s² 2p⁵

- b) Al
- c) V
- d) Cur
- e) Kr

d) Rn

Write <u>noble gas</u> electron configurations for the following atoms: 3. How many core, outer, and valence electrons do each of the following atoms have?

Core Outer Valence

11

- a) Cd [Kr] 5s² 4d¹⁰
 b) 1
 c) Ba
- a) H 0 1 1 b) Si 10 4 4 c) Ti 18 2 4 d) Br 28 7 7

36

e) Ag

am exceeption):

4. Write orbital notations for the following atoms:

a) O	11	11	11 1 1	-	7 <u>200</u> 7220720	(122)	
100	1 s	2s	2p	3s	3р	4s	3d

c) Cr
$$\frac{1}{\text{Is}}$$
 $\frac{1}{2\text{s}}$ $\frac{1}{2\text{p}}$ $\frac{1}{3\text{s}}$ $\frac{1}{3\text{p}}$ $\frac{1}{4\text{s}}$ $\frac{1}{2}$ $\frac{1}$

d) Co
$$\frac{11}{1s}$$
 $\frac{11}{2s}$ $\frac{11}{2p}$ $\frac{11}{3s}$ $\frac{11}{3p}$ $\frac{11}{3p}$ $\frac{11}{4s}$ $\frac{11}{11}$ $\frac{1}{3d}$ $\frac{1}{1}$

Electron Configuration Problems

SHETTY, M.N.

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potential barriers It provides solutions to the problems that how the energy levels of a free atom lead to the formation of energy bands in solids The statistics of the distribution of particles in different energy states in a solid has been detailed leading to the derivation of Maxwell Boltzmann Bose Einstein and Fermi Dirac statistics and their mutual relationships Quantitative derivation of the Fermi energy has been obtained by considering free electron energy distribution in solids and then considering Fermi Dirac distribution as a function of temperature The derivation of the Richardson's equation and the related work function has been quantitatively dealt with The phenomenon of tunnelling has been dealt with in terms of quantum mechanics whereas the band structure and electronic properties of materials are given quantitative treatment by using Fermi Dirac distribution function Section 2 deals with the nature of the chemical bonds types of bonds and their effect on properties followed by a detailed presentation of crystal structures of some common materials and a discussion on the structures of C60 and carbon nanotubes Coordination and packing in crystal structures are considered next followed by a detailed X ray analysis of simple crystal structures imperfections in crystals diffusion phase equilibria and mechanical behaviour Section 3 deals with thermal and electrical properties and their mutual relationships Calculations of Debye frequency Debye temperature and Debye specific heat are presented in great detail A brief section on superconductivity considers both the conventional and the high TC superconductors Sections 4 and 5 deal with the magnetic and dielectric materials considering magnetic properties from the point of view of the band theory of solids Crystal structures of some common ferrites are given in detail Similarly the displacement characteristics in dielectrics are considered from their charge displacements giving rise to some degree of polarization in the materials **Concepts And Problems In Physical** Chemistry P.S. Raghavan, 1997 Contents Introduction Atoms Molecules and Formulas Chemical Equations and Stoichiometry Aqueous Reactions and Solution Stoichiometry Gases Intermolecular Forces Liquids and Solids Atoms Structure and the Periodic Table Chemical Bonding Chemical Thermodynamics Solutions Chemical Kinetics Chemical Equilibrium Acids and Bases Ionic Equilibria I Ionic Equilibria II Redox Reactions Electrochemistry Nuclear Chemistry

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