

8 • Electron Configurations & Periodicity

WRITING ELECTRON CONFIGURATIONS

For each given element, fill in the orbital diagram and then write the electron configuration for the element.

1.	2.	3.	4.	5.	6.
Element: Ar # of e ⁻ s: <u>18</u>	Element: Mg # of e ⁻ s: <u>12</u>	Element: N # of e ⁻ s: <u>7</u>	Element: Li # of e ⁻ s: <u>3</u>	Element: P # of e ⁻ s: <u>15</u>	Element: Cl # of e ⁻ s: <u>17</u>

Write the electron configurations of each of these in **long form** and **short form**:

- Ar $1s^2 2s^2 2p^6 3s^2 3p^6$
 Ar $[\text{Ne}] 3s^2 3p^6$
- Mg $1s^2 2s^2 2p^6 3s^2$
 Mg $[\text{Ne}] 3s^2$
- N $1s^2 2s^2 2p^3$
 N $[\text{He}] 2s^2 2p^3$
- Li $1s^2 2s^1$
 Li $[\text{He}] 2s^1$
- P $1s^2 2s^2 2p^6 3s^2 3p^3$
 P $[\text{Ne}] 3s^2 3p^3$
- Cl $1s^2 2s^2 2p^6 3s^2 3p^5$
 Cl $[\text{Ne}] 3s^2 3p^5$



- Fill in the orbital diagram for the element, Fe, and write the electron configuration of Fe in the long and short form.



Electron Configuration Practice Orbital Answers

Brendan G. Carr

A red circular graphic with a gradient, appearing as a partial circle or a stylized 'C' shape, located to the right of the author's name.

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In a digitally-driven world wherever screens reign great and quick connection drowns out the subtleties of language, the profound strategies and psychological subtleties concealed within phrases frequently get unheard. Yet, situated within the pages of **Electron Configuration Practice Orbital Answers** a fascinating literary value pulsing with raw emotions, lies a fantastic journey waiting to be undertaken. Composed by an experienced wordsmith, this charming opus invites viewers on an introspective journey, lightly unraveling the veiled truths and profound impact resonating within ab muscles fabric of every word. Within the emotional depths of this moving review, we shall embark upon a sincere exploration of the book is key styles, dissect its interesting writing style, and succumb to the effective resonance it evokes strong within the recesses of readers hearts.

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