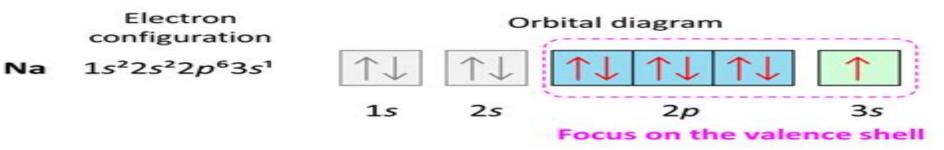
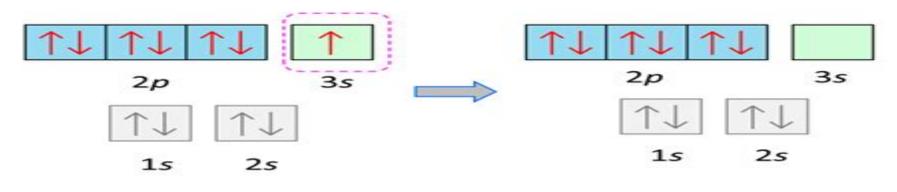
# **Electron Configuration of Main Group Cations**



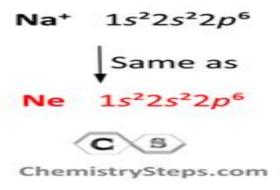
The charge is +1, therefore, one electron is lost:

The electron must be from the outermost valence orbital  $-3s^1$ 



Na  $1s^2 2s^2 2p^6 3s^1$ 

Cations of main group metals attain the electron configuration of the noble gas that precedes them in the periodic table. They are *isoelectronic* with the previous noble gas.



# **Electron Configuration Of Atoms And Ions Lab**

**M Planty** 

**Electron Configuration Of Atoms And Ions Lab:** Chemistry in the Laboratory James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations **Chemistry** Dorin, Prentice Hall Dictionary Editors, 1992 Nuclear Science Abstracts, 1974 2024-25 NVS Lab Attendant/Assistant Solved Papers YCT Expert Team, 2024 25 NVS Lab Attendant Assistant Solved Papers 592 995 Bilingual E This book contains previous year solved papers 66 sets and 5875 objective questions **Transplutonium Elements** Charles E. Stuber,1972 Nuclear Scientific and Technical Aerospace Reports ,1994 Science Abstracts ,1972 Energy Research Abstracts ,1984 ERDA Energy Research Abstracts United States. Energy Research and Development Administration, 1977 **ERDA Energy Research Abstracts** United States. Energy Research and Development Administration. Technical Information Center, 1977 Energy Research Abstracts ,1994 U.S. Government Research & Development Reports ,1970 **AFOSR.** United States. Air Force. Office of Scientific Research, 1968 Materials Chemistry Bradley D. Fahlman, 2023-03-12 This award winning textbook delivers an earnest and comprehensive treatment of the rapidly evolving

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features a revised art programme and explains the rational for each experiment

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