

## Chapter

# 10

# ELECTROCHEMISTRY

## MULTIPLE CHOICE QUESTIONS

- Reduction always takes place: (LHR 05, 12)
  - At anode
  - At cathode
  - At both electrodes
  - does not occur at electrodes
- The electrolyte KOH is used in cell: (GRW 06)
  - Lead accumulator
  - Ni – Cd cell
  - Alkaline battery
  - Silver oxide battery
- In the reaction  $2\text{Fe} + 3\text{Cl}_2 \rightarrow 2\text{FeCl}_3$ : (BWP 10, GRW 09, LHR 06)
  - Fe is reduced
  - Fe is oxidized
  - $\text{Cl}_2$  is oxidized
  - None of these happens
- Oxidation state of hydrogen in  $\text{CaH}_2$  is: (GRW 07)
  - +1
  - 1
  - +2
  - zero
- Cu metal can be purified in electrolytic cell by making the impure Cu as: (GRW 07)
  - Anode
  - Cathode
  - Anode and Cathode
  - Depends upon nature of solution
- Loss of electrons is called: (LHR 07)
  - Oxidation
  - Reduction
  - Hydration
  - Dehydration
- Fuel cells convert chemical energy into: (GRW 07)
  - Heat energy
  - Light energy
  - Electrical energy
  - Mechanical energy
- Electrolysis is used for: (LHR 08)
  - Electroplating
  - Manufacture of sodium metal
  - Manufacture of Al
  - All of these
- Nelson's cell and Down's cell are example of: (FSD 09, GRW 08)
  - Electrochemical cell
  - Galvanic cell
  - Electrolytic cell
  - None of these
- The oxidation number of Cr in  $\text{K}_2\text{Cr}_2\text{O}_7$  is: (GRW 09, 10, LHR 10, 12, 13)
  - +14
  - +12
  - +6
  - +13
- The electrolyte used in fuel cell is: (LHR 09, DGK, GRW 10)
  - Aqueous  $\text{NaCl}$
  - Molten  $\text{NaCl}$
  - $\text{KOH}$
  - $\text{H}_2\text{SO}_4$

# Electrochemistry Multiple Multiple Choice Questions

**Marcel A. Müller**



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