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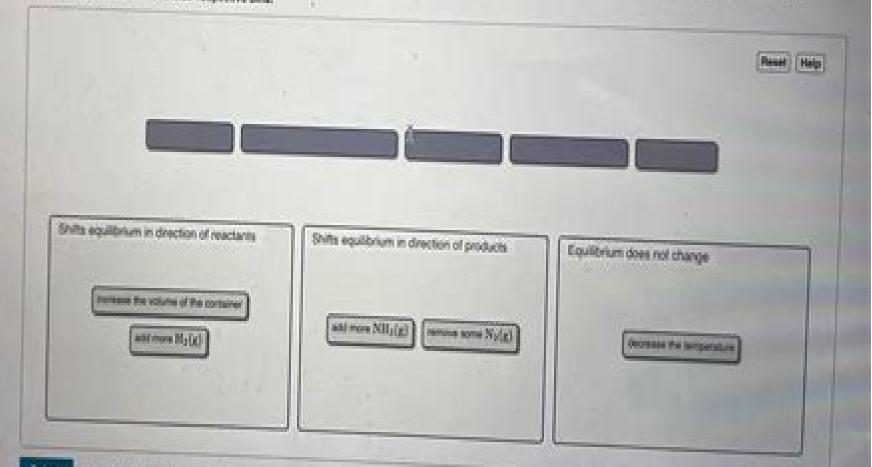
## Problem 13.36 - Enhanced - with Feedback

Ammonia is produced by reacting nitrogen gas and hydrogen gas is the following reaction:

$$N_1(g) + 3H_2(g) := 2NH_2(g) + 92 \; kJ$$

For each of the following changes at equilibrium, indicate whether the equilibrium shifts in the direction of products, shifts in the direction of reactants, or does not change.

Drag the appropriate items to their respective bins.



# **Mastering Chemistry Answers Ch 13**

**Andrew Case** 

#### **Mastering Chemistry Answers Ch 13:**

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