



Exercise 10.33

The following figures show several molecular geometries.



Tap image to zoom

Part A

Give the number of total electron groups, the number of bonding groups, and the number of lone pairs for (a) geometry.

Express your answers as an integer. Enter your answers in ascending order separated by commas.

$N_{\text{total}}, N_{\text{bonding}}, N_{\text{lone}} =$

Submit

Give Up

Part B

Give the number of total electron groups, the number of bonding groups, and the number of lone pairs for (b) geometry.

Express your answers as an integer. Enter your answers in ascending order separated by commas.

$N_{\text{total}}, N_{\text{bonding}}, N_{\text{lone}} =$

Mastering Chemistry Exercise Answers

Marti Anne Maguire, Paula Stiles



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