#### Endothermic and Exothermic reaction Worksheet

Name \_\_\_\_\_ date\_\_\_\_ period \_\_\_\_

#### 1 Exothermic and endothermic reactions

Decide whether each of these reactions is exothermic or endothermic:

- When two chemicals mix their temperature rises: \_\_\_\_exo\_\_\_\_\_\_
- A solid burns brightly and releases heat, light and sound: \_\_\_\_\_exo\_\_\_\_\_
- c) When two chemicals are mixed their temperature drops: \_\_\_\_\_endo\_\_\_\_\_
- d) Two chemicals will only react if you heat them continually: endo endo
- e) Plants take in light energy for photosynthesis: \_\_\_\_endo\_\_\_\_

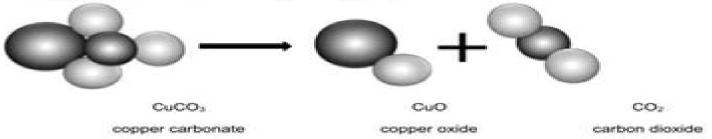
## Making and breaking bonds

During chemical reactions the bonds between atoms break and new bonds form.

Energy must be absorbed to break a bond, so breaking bonds is endothermic.

Making new bonds is exothermic because energy is released.

a) When green copper carbonate decomposes, the equation is:



Is the reaction exothermic or endothermic? Use ideas about bonds to explain why.

\_exo because two bonds are broken from 4 bonds

b) Draw diagrams to show what happens when hydrogen reacts with oxygen. Mark the bonds broken in blue and the new bonds formed in red. The equation is:





2H<sub>2</sub>Ored side water

# **Key To Radioactivity And Nuclear Reactions Answers**

**LP Steffe** 

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