

Types of Nuclear Radiation – Questions

Instructions: Identify the following descriptions as alpha, beta, or gamma:

- | | |
|--|--------------|
| 1) This radiation can only be stopped by <u>thick concrete or lead</u> . | <u>Gamma</u> |
| 2) This particle can be stopped by <u>clothing alone</u> . | <u>Alpha</u> |
| 3) During this decay, <u>no protons, neutrons, or electrons are lost</u> . | <u>Gamma</u> |
| 4) During this type of decay, <u>two protons</u> are lost. | <u>Alpha</u> |
| 5) This particle has a <u>-1 charge</u> . | <u>Beta</u> |
| 6) This particle has a mass of 4. | <u>Alpha</u> |
| 7) This type of radiation is identical to an <u>electron</u> . | <u>Beta</u> |
| 8) This form of radiation has <u>no mass</u> and <u>no charge</u> . | <u>Gamma</u> |
| 9) This particle is represented as a <u>helium</u> atom. | <u>Alpha</u> |
| 10) This particle is emitted as in the form of <u>waves</u> . | <u>Gamma</u> |
| 11) This particle can be stopped by a <u>piece of paper</u> . | <u>Alpha</u> |
| 12) This particle has average (medium) penetrating power. | <u>Beta</u> |
| 13) This is the <u>least dangerous</u> type of radiation. | <u>Alpha</u> |
| 14) This is the <u>most dangerous</u> type of radiation. | <u>Gamma</u> |

Instructions: Use the nuclear equation below to answer the following questions:



- | | |
|---|--------------------|
| 15) The picture above represents <u>what type</u> of nuclear decay? | <u>Beta Decay</u> |
| 16) What is the mass of the starting phosphorus (P) atom? | <u>32</u> |
| 17) What is the mass of the particle released above? | <u>0</u> |
| 18) Which decay would have caused a larger change in mass? | <u>Alpha Decay</u> |

Key Terms Radioactivity And Nuclear Reactions

Worksheet Answer

Roman Wölfel



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