

ঘের ও কীটপতঙ্গের অংশের নির্ধারিত পটীকার প্রস্তুতকরণ

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} = 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

$$W = 20 \text{ g}$$

$$HCl \text{ এর আপেক্ষিক ভর,}$$

$$M = 36.5 \text{ g/mol}$$

$$\text{চলকের আয়তন, } V = 1 \text{ L}$$

$$\text{চলকের ঘনমাত্রা } S = 1$$

উদাহরণে উল্লিখিত N_2 -পদ্ধতিতে 10 mg এর মধ্যে পদার্থ পরিমাণ মূল ও V_1 -পদ্ধতিতে 0.2 g 2% $\left(\frac{W}{V}\right)$ HCl নমুনা নিয়ে কাজ করা হয়েছে।

যদিও N_2 -পদ্ধতিটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও V_1 -পদ্ধতিটি হলো মাইক্রো আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব। নিম্নে তা বিশ্লেষণ করা হলো-

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় প্রায় 100 ml রাসায়নিক উপাদান সন্নিবেশিত ও বিশ্লেষণের বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে মৃদু গন্ধ উৎপাদনযোগ্যভাবে কমে আসে।

এছাড়া সেমি-মাইক্রো পদ্ধতিতে H_2SO_4 এর পরিমাণে কয়েকটিমাইক্রো ব্যবহার করা হয় ফলে পরিবেশে মৃদু গন্ধ উঠে না।

$(CH_3NH_2 + H_2O \rightarrow CH_3NH_3^+ + OH^-)$ ফলে কোনো কোনো ক্ষেত্রে কম কষ্টিকর রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে মৃদু গন্ধ সঞ্চারিত কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

চিত্র-১০



১. উদাহরণে চলকের কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব কেন?
৩. চলকের জন্য প্রয়োজ্য হ্যান্ডলার প্রকৌশলটি অঙ্কন করে এবং ব্যাখ্যা করে।
৪. উদাহরণের অংশের চলক থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে 1M ঘনমাত্রার চলক তৈরি করার প্রক্রিয়া ব্যাখ্যা করে। ৪ নং প্রশ্নের উত্তর

৫. বিশ্লেষণের নিয়ন্ত্রিত কক্ষের উপর রাইজার স্থাপন করলে কক্ষের প্রতি দিকের দৃশ্য যে ভর পাওয়া যায়, তাকে রাইজার দৃশ্যক বলে।

৬. সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় প্রায় 100 ml রাসায়নিক উপাদান সন্নিবেশিত ও বিশ্লেষণের বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে মৃদু গন্ধ উৎপাদনযোগ্যভাবে কমে আসে।

৭. উদাহরণের চলকটি হলো সালফিউরিক এসিড (H_2SO_4) এর 10 M চলক। H_2SO_4 একটি জারক ও ক্ষয়কারী পদার্থ। H_2SO_4 এর 10 M চলক মানে পাত্র H_2SO_4 চলক। পাত্র H_2SO_4 পদার্থকে ক্ষয়কারী হ্যান্ডলার প্রকৌশল অঙ্কন করা হয়। পাত্র H_2SO_4 এর জন্য ক্ষয়কারী হ্যান্ডলার প্রকৌশল অঙ্কন করে ব্যাখ্যা করা হলো:

| প্রকৌশল | ক্ষয়কারী (Corrosive) |
|------------------|--|
| আয়তন | ক্ষয়কারী রাসায়নিক উপাদান চলক যে ক্ষয়কারী রাসায়নিক দ্রব্যকে ক্ষয়কারী হ্যান্ডলার প্রকৌশল অঙ্কন করা হয়। এতে খুব খুঁচু হ্যান্ডলার এবং অন্যদিকে পদার্থ পড়ে যায়। |
| উপাদান | পাত্র কঠিন এসিড/বাস (HCl, HNO ₃ , H ₂ SO ₄) পাত্র চলক (NaOH, KOH, H ₂ O ₂) আয়তন/লিটার প্রকৌশল চলক ইত্যাদি। |
| সংরক্ষণ পাত্র/ভা | <ul style="list-style-type: none"> পাত্র এসিডকে পোড়ালে ক্ষয়কারী হবে। পাত্র ক্ষয়কারী, এখানে ক্ষয়কারী পদার্থকে পোড়ালে ক্ষয়কারী হবে। |

৮. উদাহরণের চলকটি হলো 10M H_2SO_4 চলক এবং এর থেকে 1M ঘনমাত্রার চলক প্রস্তুত করতে হবে 1M ও এর কাছাকাছি ব্যবহার করতে হবে।

1M কাছাকাছি হলো 100 ml আয়তন/লিটার চলক ও এর টি হলো পিপিটি। সেমি-মাইক্রো আয়তন/লিটার চলকের আয়তন 100 ml, তাই আমাদের 100 ml 1M ঘনমাত্রার H_2SO_4 চলক তৈরি করতে হবে। চলকের মৃদু গন্ধ খুব হতে জানে না।

$$V_1 S_1 = V_2 S_2$$

$$\text{যা, } V_1 = \frac{V_2 S_2}{S_1}$$

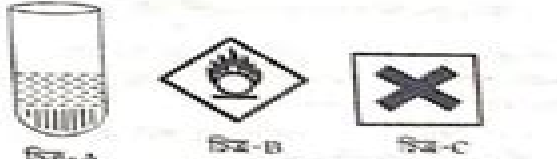
$$= \left(\frac{100 \times 1}{10} \right) \text{ mL}$$

$$= 10 \text{ mL}$$

অতএব, 100ml H_2SO_4 চলকের সঙ্গে পানি যোগ করতে হবে (100-10) mL = 90 mL।

আবার, পিপিটির সাহায্যে 10 mL চলক ঘনমাত্রার চলক নিয়ে তার সাথে আয়তন/লিটার চলকে 90 mL পানি যোগ করলে আমাদের কাঙ্ক্ষিত ঘনমাত্রার চলক পাওয়া যাবে।

চিত্র-১১



১. অ্যান্টিঅক্সিডেন্ট কাকে বলে?
২. সামান্য চলক K₂ এর মানে কখনো অসীম হয় না-ব্যাখ্যা করে।
৩. A চিত্রের হ্যান্ডলারের উত্তর করার কৌশল বর্ণনা করে।
৪. B ও C চিত্রের সাথে সম্পর্কিত রাসায়নিক চলকটির ব্যবহারকালে খাখা কৃষ্ণি হলে কবীর উপর সম্পর্কিত বিশ্লেষণ করে। ৪ নং প্রশ্নের উত্তর

৫. অ্যান্টিঅক্সিডেন্ট এক ধরনের রাসায়নিক বা কৃত্রিম দ্রব্য সঞ্চারক যা মূলত খাদ্য উপাদানসমূহকে জারণ প্রতিরোধ করে এবং ক্ষয় করে।

৬. সামান্য চলক K₂ এর মানে কখনো অসীম বা শূন্য হতে পারে না। উদাহরণ ব্যাখ্যা করার জন্য নিম্নের উদাহরণ মাইক্রোপট বিশ্লেষণ করি-
 $A + B \rightarrow C + D$

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