

- A. A rug that really brings the room together
- B. A cat
- C. A dog
- D. A TV

14. When the narrator is in Myrtle's Manhattan apartment, why is the narrator fuzzy on what happened that afternoon?

- A. He doesn't understand what he saw
- B. There were so many people that it was all a blur
- C. He had a concussion from being punched by Tom
- D. He was drunk

15. When the narrator is in Myrtle's Manhattan apartment, what rumor does the narrator hear about Gatsby?

- A. That he isn't as rich as he seems
- B. That he's a cousin of Kaiser Wilhelm
- C. That he killed a man
- D. That he's involved in bootleg liquor

16. When the narrator is in Myrtle's Manhattan apartment, according to Catherine, why can't Myrtle and Tom get married?

- A. Tom still loves Daisy
- B. Daisy is Catholic and doesn't believe in divorce
- C. Myrtle and Tom don't really love each other
- D. Myrtle can't afford to divorce George

17. What caused Myrtle to regret marrying George?

- A. When she found out he borrowed a suit for their wedding
- B. When George's business failed
- C. She met Tom
- D. When she had their first child

18. During the party in Myrtle's Manhattan apartment, what does Tom do when Myrtle says Daisy's name?

- A. He promises to divorce Daisy
- B. He kicks her out of the apartment
- C. He breaks Myrtle's nose
- D. He leaves without saying a word

19. When the narrator leaves the party in Myrtle's Manhattan apartment, what happens to him after he gets into the elevator with Mr. McKee?

- A. He lets Mr. McKee take a portrait of him
- B. He calls the police
- C. He takes the subway to Central Park
- D. He has a "romantic encounter" with Mr. McKee

Great Gatsby Multiple Choice Test And Answers

Jeff Halstead



Great Gatsby Multiple Choice Test And Answers:

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Modern Classroom Assessment Bruce B. Frey,2013-09-13 Modern Classroom Assessment offers an applied student centered guide to the major research based approaches to assessment in today s modern classroom Rather than simply list basic assessment formats with a few examples as many textbooks do award winning professor and scholar Bruce Frey s book fully explores all five key approaches for teacher designed assessment Traditional Paper and Pencil Performance Based Assessment Formative Assessment Universal Test Design and Authentic Assessment while making abstract concepts and guidelines clear with hundreds of real world illustrations and examples of what actual teachers do Offering a variety of engaging learning tools and realistic stories from the classroom this text will give any reader a strong foundation for designing modern assessments in their own classrooms

My Max Score AP English Literature and Composition Tony Armstrong,2011 Provides test taking tips and strategies reviews topics on the test and includes a full length practice exam with answers and explanations

New Directions in Assessing Historical Thinking Kadriye Ercikan,Peter Seixas,2015-02-20 New technologies have radically transformed our relationship to information in general and to little bits of information in particular The assessment of history learning which for a century has valued those little bits as the centerpiece of its practice now faces not only an unprecedented glut but a disconnect with what is valued in history education More complex processes historical thinking historical consciousness or

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Excellence Jeff Halstead, 2014-07-02 Empowering Excellence inspires a revolution in teaching and assessment practices These methods of teaching and grading represent a major step ahead in how teachers assess and interact with students Empowering Excellence presents 12 significant shifts that encourage motivation build confidence and measure what students know and are able to do based on the Common Core State Standards or equivalent standards in a highly supportive student centered environment Teachers administrators and education professors will find ideas that will transform classrooms into places where students are filled with confidence and empowered to strive for excellence **Call of the Wild, The:**

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of Effective Teachers James H. Stronge, 2018-03-21 Every teacher seeks to be an effective teacher Every teacher wants to have a positive remarkable and lasting influence on students lives But what makes for an effective teacher What role does teacher preparation play in teacher effectiveness What do effective teachers do during planning instruction and assessment How do they create a learning environment that engages and supports students And how do effective teachers interact with their students to promote the best opportunities and results for all In Qualities of Effective Teachers 3rd edition James H Stronge explores these questions and more as he synthesizes the literature on teacher effectiveness The result A research based framework for effective teaching that addresses Professional knowledge Instructional planning Instructional delivery Assessment Learning environment Professionalism Stronge also examines characteristics of effective teachers of at risk students and high ability students To bridge the gap between research and practice he includes checklists of skills and positive qualities associated with effective teacher performance as well as red flags that indicate that teachers may not be reaching their full potential in the classroom This resource is for any educator interested in improving teaching It offers

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web purpose in this experiment you will observe a single replacement reaction you will also use stoichiometry to predict what the theoretical yield of product is and calculate a percent yield reaction iron copper ii sulfate copper iron ii sulfate pre lab questions write the balanced equation

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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate CuSO_4 and 2 0153 grams of iron powder Fe the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the

equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web under the finalise of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away copper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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