Time : 3.00 Hrs

Marks: 70

#### PART - I

- Answer all the questions.  $15 \times 1 = 15$ T
- Which of the following is / are true with respect to Carbon 12 1. a) relative atomic mass is 12u b) oxidation number of Carbon is +4 in all its compounds c) mole of Carbon - 12 contain 6.022X1022 carbon atoms d) all of these
- How many electrons in an atom with atomic number 105 can have (n + I) = 82... b) 17 d) unpredictable
- Among the following which is isoelectronic with O2- ion. 3. d) N2c) Ca2+ D) F a) C/-
- The cause of permanent hardness of water is due to 45 ... b) Mg (HCO3)2 c) CaC/d) Macon a) Ca(HCO<sub>3</sub>)<sub>2</sub>
- Sodium is stored in 55. b) Water c) Kerosene d) None of these. a) Alcohol
- It temperature and volume of an ideal gas is increased to twice its values, the initial 6. d) 3p pressure P becomes a) 4p b) 20 c) p
- 7... Heat of combustion is always d) either positive or negative a) Positive b) negative: c) zero
- In the reaction, Fe(OH)3(s) == Fe3+(aq) + 3OH-(aq) if the concentration of OH-ions 8. is decreased by 1/4 times, then the equilibrium concentration of Fe3+ will b) also decreased by 1/4 times a) not changed d) increase by 64 times c) increase by 4 times
- According to Raoults law, the relative lowering of vapour pressure for a solution is equal 9. to a) mole fraction of solvent b) mole fraction of solute
  - d) number of moles of solvent c) number of moles of solute
- Number of sigma (a) and pi (x) bonds in acetylene is ....... 10. d) 30 & 2m b) 3 or & 3 n c) 2 o & 3 m a) 2\sigma 8 2\pi
- 11. Sodium nitropruside reacts with sulphide ion to give a purple colour due to the formation
- a) [Fe(CN)5 NO]3- b) [Fe (NO)5 CN]+ c) [Fe(CN)5 NOS]4- d) [Fe(CN)5 NOS]3-
- What is the hybridisation state of benzyl carbonium ion? 12. d) sp<sup>2</sup>db) spd<sup>2</sup>  $c) sp^3$ a) sp<sup>2</sup>
- ...... is used for welding and cutting metals. 13. a) Ethylene b) Propylene c) Poly ethylene d) Oxy - acetylene
- What should be the correct IUPAC name of diethyl chloromethane? 1.4. a) 3 - chloro pentane b) 1 - chloropentane c) 1 - chloro - 1, 1, diethyl methane d) 1 - chloro - 1 - ethylepropane
- The pH of normal rain water is ...... 15. d) 4.6 a) 6.5b) 7.5

#### PART - II

- Answer any six. (q.No. 24 is compulsory). II
- 16. Define modern periodic law.
- 17. What is water - gas shift reaction?

- $6 \times 2 = 12$
- HSS 11 வேதியியல் EM PAGE 1

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**Ian Pickup** 

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