

## The Periodic Table

H																	He
Li	Be											B	C	N	O	F	Ne
Na	Mg											Al	Si	P	S	Cl	Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	Ac	Rf	Ha	Sg	Uns	Uno	Une									
		Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu		
		Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr		

- The elements in periods 2 through 7 that are to the left of the zigzagged line are metals. List five metals.  
\_\_\_\_\_
- The elements in periods 1 through 6 that are to the right of the zigzagged line are nonmetals. List five nonmetals.  
\_\_\_\_\_
- The elements that border either side of the zigzagged line are metalloids. Metalloids have properties of both metals and nonmetals. Name two metalloids.  
\_\_\_\_\_
- The alkali metals are in group 1. They have only one electron in their outer shell. They are very reactive and have a low melting point. Shade the alkali metals blue.
- The noble gases are found in group 18. Their outer energy level is filled, and they are very nonreactive, colorless gases. Shade the noble gases red.
- Halogens have seven electrons in the outer shell. They are located in group 17 on the periodic table. Halogens react with metals to form salts. Shade the halogens green.
- The transition elements are found in groups 3 through 12 and periods 4 through 7. These elements have either one or two electrons in the outer energy level. They are often used to form alloys because they are hard and have high melting points. Shade the transition elements orange.
- Alkaline earth metals are located in group 2. They have two electrons in their outer energy level. Shade the alkaline metals purple.

# Exploring The Periodic Table Skills Worksheet Answers

**Raffaela Di Napoli**



## Exploring The Periodic Table Skills Worksheet Answers:

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Crete in the summer of 1992 The purpose of the Advanced Study Institute was to bring together a small number of senior lecturers and advanced graduate students to investigate and discuss the psychological and educational foundations of technology based learning environments and to draw the implications of recent research findings in the area of cognitive science for the development of educational technology As is apparent from the diverse nature of the contributions included in this volume the participants at the ASI came from different backgrounds and looked at the construction of technology based learning environments from rather diverse points of view Despite the diversity a surprising degree of overlap and agreement was achieved Most of the contributors agreed that the kinds of technology supported learning environments we should construct should stimulate students to be active and constructive in their knowledge building efforts embed learning in meaningful and authentic activities encourage collaboration and social interaction and take into consideration students prior knowledge and beliefs

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