

Endothermic and Exothermic reaction Worksheet

Name _____ date _____ period _____

1 Exothermic and endothermic reactions

Decide whether each of these reactions is exothermic or endothermic:

- a) When two chemicals mix their temperature rises: ____exo____
- b) A solid burns brightly and releases heat, light and sound: ____exo____
- c) When two chemicals are mixed their temperature drops: ____endo____
- d) Two chemicals will only react if you heat them continually: ____endo____
- e) Plants take in light energy for photosynthesis: ____endo____

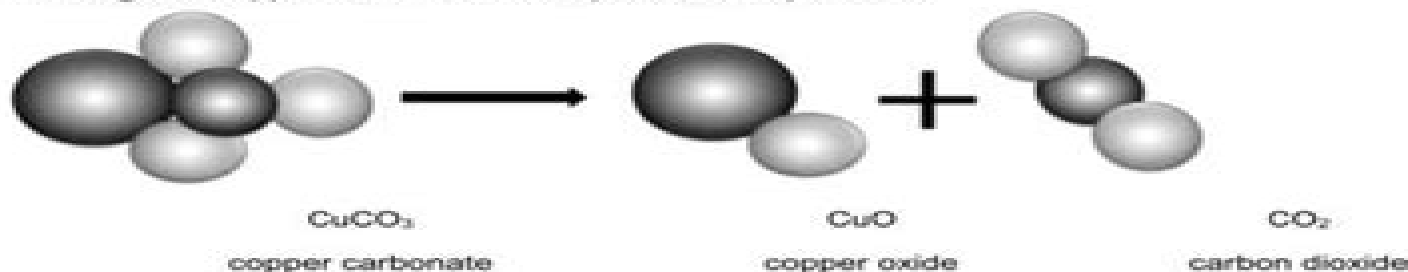
2 Making and breaking bonds

During chemical reactions the bonds between atoms break and new bonds form.

Energy must be absorbed to break a bond, so breaking bonds is endothermic.

Making new bonds is exothermic because energy is released.

- a) When green copper carbonate decomposes, the equation is:



Is the reaction exothermic or endothermic? Use ideas about bonds to explain why.

__exo because two bonds are broken from 4 bonds

- b) Draw diagrams to show what happens when hydrogen reacts with oxygen. Mark the bonds broken in blue and the new bonds formed in red. The equation is:



Key To Radioactivity And Nuclear Reactions Answers

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Table of Contents Key To Radioactivity And Nuclear Reactions Answers

1. Understanding the eBook Key To Radioactivity And Nuclear Reactions Answers
 - The Rise of Digital Reading Key To Radioactivity And Nuclear Reactions Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Key To Radioactivity And Nuclear Reactions Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Key To Radioactivity And Nuclear Reactions Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Key To Radioactivity And Nuclear Reactions Answers
 - Personalized Recommendations
 - Key To Radioactivity And Nuclear Reactions Answers User Reviews and Ratings
 - Key To Radioactivity And Nuclear Reactions Answers and Bestseller Lists
5. Accessing Key To Radioactivity And Nuclear Reactions Answers Free and Paid eBooks
 - Key To Radioactivity And Nuclear Reactions Answers Public Domain eBooks
 - Key To Radioactivity And Nuclear Reactions Answers eBook Subscription Services
 - Key To Radioactivity And Nuclear Reactions Answers Budget-Friendly Options

6. Navigating Key To Radioactivity And Nuclear Reactions Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Key To Radioactivity And Nuclear Reactions Answers Compatibility with Devices
 - Key To Radioactivity And Nuclear Reactions Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Key To Radioactivity And Nuclear Reactions Answers
 - Highlighting and Note-Taking Key To Radioactivity And Nuclear Reactions Answers
 - Interactive Elements Key To Radioactivity And Nuclear Reactions Answers
8. Staying Engaged with Key To Radioactivity And Nuclear Reactions Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Key To Radioactivity And Nuclear Reactions Answers
9. Balancing eBooks and Physical Books Key To Radioactivity And Nuclear Reactions Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Key To Radioactivity And Nuclear Reactions Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Key To Radioactivity And Nuclear Reactions Answers
 - Setting Reading Goals Key To Radioactivity And Nuclear Reactions Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Key To Radioactivity And Nuclear Reactions Answers
 - Fact-Checking eBook Content of Key To Radioactivity And Nuclear Reactions Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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