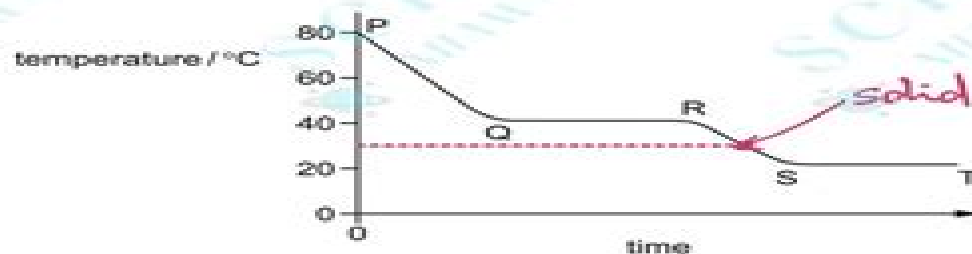


- 1 Substance M is a solid at 30°C .

The substance is heated to 80°C and its temperature measured as it cools down to room temperature.

The cooling curve is shown.



Between which times is substance M freezing?

- A P to Q **B** Q to R C R to S D S to T

- 2 Which gas has the fastest rate of diffusion?

- A Ar **B** C_2H_6 C HCl D H_2S
- Handwritten notes:* 40, $24 + 6 = 30$, 36.5, 34

- 3 There are two stable isotopes of bromine.

The mass number of isotope 1 is 79.

The mass number of isotope 2 is 81.

Which statement is correct?

- A The isotopes have the same number of neutrons.
 B The isotopes have different chemical properties.
 C The isotopes have different numbers of protons.
D The isotopes have the same number of outer electrons.

- 4 Which statement about ions and ionic bonds is correct?

- A** Bromine atoms form negatively charged bromide ions.
 B Ionic bonds form between elements in Group VII of the Periodic Table.
 C Positive ions are formed when atoms lose protons.
 D Potassium iodide contains negatively charged potassium ions.

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Marcel A. Müller



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