Q.ID

Marks: 80 Marks Time: 3Hrs.

General Instructions:

i. This question paper consists of 39 questions in 5 sections.

ii. All questions are compulsory. However, an internal choice is provided in some questions. A student is expected to attempt only one of these questions.

iii Section A consists of 20 objective type questions carrying I mark each.

iv. Section B consists of 6 very short questions carrying 02 marks each. Answers to these questions should be in the range of 30 to 50 words.

 Section C consists of 7 short answer type questions carrying 03 marks each. Answers to these questions should be in the range of 50 to 80 words.

vi. Section D consists of 3 long answer type questions carrying 05 marks each. Answer to these questions should be in the range of 80 to 120 words.

vii. Section F. consists of 3 source-based case-based units of assessment of 04 marks each with sub-parts.

Section-A		
Select and write the most appropriate option out of the four options given for each of the questions 1 - 20.	1N20 = 20M	
Given the reaction:		
$Zn+CuSO_4\rightarrow ZnSO_4+Cu$	1	
which of the following statements is true? a) Zinc is reduced and copper is oxidised.		

b) Zine is reduced and copper is oxidised.
b) Zine is oxidised and copper is reduced.
c) Both zine and copper are oxidised.
d) Both zine and copper are reduced.

Q.2) The reaction between MnO₂ with HCl is depicted in the following diagram. It was observed that a gas with bleaching abilities was released.

leaching abilities was released.



The chemical reaction between above mentioned reactants:

a) displacement reaction

b) combination reaction

c) redox reaction

d) decomposition reaction.

(23) Identify 'x', 'y' and 'z' in the following balanced reaction

$$xPb(NOz)z(s) \xrightarrow{Heat} yPbO(s) + zNOz(g) + Oz(g)$$

- (a) 2, 4, 2
- (b) 2, 2, 4
- (c) 2, 4, 4

(d) 4, 2, 2

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Christian G. Meyer

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