



Introduction to working as a Chemist.

Use a highlighter as you work through this booklet. ESL students, make vocab lists of the terminology that is defined or described here.

Assessment Statements

- | | |
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| <ul style="list-style-type: none"> <input type="checkbox"/> Outline the Scientific Method <input type="checkbox"/> Identify independent, dependent & controlled variables. <input type="checkbox"/> Demonstrate correct and safe use of laboratory equipment. <input type="checkbox"/> Distinguish between quantitative and qualitative data <ul style="list-style-type: none"> ➢ <i>Qualitative is descriptive, not using numbers, e.g. properties, colour, gas production, sound, odor.</i> ➢ <i>Quantitative is mathematical, including</i> <ul style="list-style-type: none"> ◊ <i>Raw data (as recorded)</i> ◊ <i>Processed (after calculations)</i> <input type="checkbox"/> Distinguish between accuracy and precision <ul style="list-style-type: none"> ➢ <i>Accuracy describes the correctness of a set of measurements</i> ➢ <i>Precision describes the repeatability of a set of measurements</i> | <ul style="list-style-type: none"> <input type="checkbox"/> State SI units of measurement for <ul style="list-style-type: none"> ➢ <i>Mass: grams (g)</i> ➢ <i>Length: Metres (m), millimeters (mm)</i> ➢ <i>Concentration: molarity (M)</i> ➢ <i>Temperature: Degrees Celsius (°C)</i> ➢ <i>Volume: Litres (L) or millilitres (mL)</i> ➢ <i>Density: grams per cubic centimeter (g/cm³)</i> ➢ <i>Conductivity: Siemens (S)</i> ➢ <i>Acidity or alkalinity (pH)</i> <input type="checkbox"/> Energy: Joules (J or kJ)Determine the uncertainty of digital and analog measuring tools <ul style="list-style-type: none"> ➢ <i>Usually ± 1 of the smallest division</i> <input type="checkbox"/> Calculate the % error of a given or recorded set of values <ul style="list-style-type: none"> ➢ <i>$(\text{Measured} - \text{True}) / \text{True} \times 100$</i> <input type="checkbox"/> State values to appropriate numbers of significant figures <input type="checkbox"/> Calculate values to appropriate numbers of significant figures |
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Related Concepts:

- ☐ 01.1 Element symbols (Quia quiz: <http://www.quia.com/quiz/3246640.htm#>)
- ☐ 01.2 Accuracy, Precision, Error, Uncertainty (Quia quiz: <http://www.quia.com/quiz/3507626.htm#>)
- ☐ 01.3 SigDigs and Notation (Quia quiz: <http://www.quia.com/quiz/3504378.htm#>)

Command Terms: Be sure you understand what is being asked of you

Analyse Break down to identify key parts. Interpret information to reach conclusions.

| | |
|-----------------|----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| Apply | Use knowledge & understanding in response to a real situation or given circumstances. |
| Calculate | Obtain a numerical answer, showing relevant stages in the working. |
| Define | Give the precise meaning of a word, phrase, concept or physical quantity. |
| Describe | Give a detailed account or picture of an event, pattern, process or pattern. |
| Design | Produce a plan, simulation or model. |
| Determine | Obtain the only possible answer. |
| Discuss | Offer a considered and balanced review that includes a range of arguments, factors or hypotheses. Opinions or conclusions should be presented clearly and supported by appropriate evidence. |
| Distinguish | Make clear the differences between two or more concepts or items. |
| Evaluate | Assess the implications and limitations; make judgments about the ideas, works, solutions or ideas in relation to selected criteria. |
| Explain | Give a detailed account including reasons or causes. |
| Identify | Provide an answers from a number of possibilities. Recognise and state briefly a distinguishing fact or feature. |
| Label | Add title, labels or brief explanation(s) to a diagram or graph. |
| List | Give a sequence of answers with no explanation. |
| Outline | Give a brief account. |
| Recall | Remember or recognize from prior learning experiences. |
| State | Give a specific name, value or other brief answer without explanation or calculation. |
| Suggest/Propose | Propose a solution, hypothesis or other possible answer. |

Exploring Chemistry Laboratory Manual

**Thirumalaisamy P.
Velavancorresponding**



Exploring Chemistry Laboratory Manual:

Exploring Chemistry; Laboratory Manual Eileen Delgado Johann, Ana Ciereszko, Dorma Gottlieb, 2003-01-01

Exploring Chemistry Laboratory Experiments in General, Organic and Biological Chemistry Julie R.

Peller, 2003-04 This lab manual is organized and written to ensure that non science majors are comfortable with chemistry labs by making the experiments more applicable to students daily lives This approach also serves to make the experiments more understandable Many labs relate specifically to allied health fields

Exploring Chemistry Ana Alejandre

Ciereszko, 2013 **ChemDiscovery Lab Manual** Olga I. Agapova, 2002

Exploring General Chemistry in the

Laboratory Colleen F. Craig, Kim N. Gunnerson, 2017-01-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes

Exploring Biology Ella

Thea Smith, 1959

Exploring Physics Richard F. Brinckerhoff, Judson B. Cross, Arthur Lazarus, 1959

Green

Chemistry Laboratory Manual for General Chemistry Sally A. Henrie, 2015-03-18 Green chemistry involves designing novel ways to create and synthesize products and implement processes that will eliminate or greatly reduce negative environmental impacts Providing educational laboratory materials that challenge students with the customary topics found in a general chemistry laboratory manual this lab manual enables students to see how green chemistry principles can be applied to real world issues Following a consistent format each lab experiment includes objectives prelab questions and detailed step by step procedures for performing the experiments Additional questions encourage further research about how green chemistry principles compare with traditional more hazardous experimental methods

Chemistry Education in the ICT Age Minu Gupta Bhowon, Sabina Jhaumeer-Laulloo, Henri Li Kam Wah, Ponnadurai Ramasami, 2009-07-21 th th The 20 International Conference on Chemical Education 20 ICCE which had rd th Chemistry in the ICT Age as the theme was held from 3 to 8 August 2008 at Le Meridien Hotel Pointe aux Piments in Mauritius With more than 200 participants from 40 countries the conference featured 140 oral and 50 poster presentations th Participants of the 20 ICCE were invited to submit full papers and the latter were subjected to peer review The selected accepted papers are collected in this book of proceedings This book of proceedings encloses 39 presentations covering topics ranging from fundamental to applied chemistry such as Arts and Chemistry Education Biochemistry and Biotechnology Chemical Education for Development Chemistry at Secondary Level Chemistry at Tertiary Level Chemistry Teacher Education Chemistry and Society Chemistry Olympiad Context Oriented Chemistry ICT and Chemistry Education Green Chemistry Micro Scale Chemistry Modern Technologies in Chemistry Education Network for Chemistry and Chemical Engineering Education Public Understanding of

Chemistry Research in Chemistry Education and Science Education at Elementary Level We would like to thank those who submitted the full papers and the reviewers for their timely help in assessing the papers for publication th We would also like to pay a special tribute to all the sponsors of the 20 ICCE and in particular the Tertiary Education Commission <http://tec.intnet.mu> and the Organisation for the Prohibition of Chemical Weapons <http://www.opcw.org> for kindly agreeing to fund the publication of these proceedings **Keep Tab on the Lab**, 1960-11 **You and Your World** Paul Franz Brandwein, 1960

Library of Congress Catalogs Library of Congress, 1976 **The Well-Trained Mind** Susan Wise Bauer, Jessie Wise, 2009-05-04 If you re a parent who has decided to educate your children yourself this book is the first you should buy Washington Times The Well Trained Mind will instruct you step by step on how to give your child an academically rigorous comprehensive education from preschool through high school one that will train him or her to read to think to understand to be well rounded and curious about learning Veteran home educators Jessie Wise and Susan Wise Bauer outline the classical pattern of education called the trivium which organizes learning around the maturing capacity of the child s mind and comprises three stages the elementary school grammar stage the middle school logic stage and the high school rhetoric stage Using this theory as your model you ll be able to instruct your child in all levels of reading writing history geography mathematics science foreign languages rhetoric logic art and music regardless of your own aptitude in those subjects This newly revised edition contains completely updated ordering information for all curricula and books new and expanded curricula recommendations new material on using computers and distance learning resources answers to common questions about home education information about educational support groups and advice on practical matters such as working with your local school board preparing a high school transcript and applying to colleges **You and Science** Paul Franz

Brandwein, 1960 **Chemistry in the Laboratory** James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations **You**

and Your Resources Paul Franz Brandwein, 1960 Experimental Organic Chemistry Joaquín Isac-García, José A. Dobado, Francisco G. Calvo-Flores, Henar Martínez-García, 2015-10-30 Experimental Organic Chemistry Laboratory Manual is designed as a primer to initiate students in Organic Chemistry laboratory work Organic Chemistry is an eminently experimental science that is based on a well established theoretical framework where the basic aspects are well established but at the same time are under constant development Therefore it is essential for future professionals to develop a strong background in the laboratory as soon as possible forming good habits from the outset and developing the necessary skills to address the challenges of the experimental work This book is divided into three parts In the first safety issues in laboratories are addressed offering tips for keeping laboratory notebooks In the second the material the main basic laboratory procedures

preparation of samples for different spectroscopic techniques Microscale Green Chemistry and qualitative organic analysis are described The third part consists of a collection of 84 experiments divided into 5 modules and arranged according to complexity The last two chapters are devoted to the practices at Microscale Synthesis and Green Chemistry seeking alternatives to traditional Organic Chemistry Organizes lab course coverage in a logical and useful way Features a valuable chapter on Green Chemistry Experiments Includes 84 experiments arranged according to increasing complexity Library of Congress Catalog Library of Congress,1970 A cumulative list of works represented by Library of Congress printed cards

Core Science Lab Manual with Practical Skills for Class IX V. K. Sally, Chhaya Srivastava, Goyal Brothers Prakashan, 2019-01-01 Goyal Brothers Prakashan **Elementary Chemistry Laboratory Manual** Siobhan Shay, 2020-06-30

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