

Substance	Lewis Structure	Molar Mass	Intermolecular Attractive Force
<b>Methanol</b> $\text{CH}_3\text{OH}$ $12 + 3(1) + 16 + 1 = 32$	$\begin{array}{c} \text{H} \\   \\ \text{H} - \text{C} - \ddot{\text{O}} - \text{H} \\   \\ \text{H} \end{array}$	32.042	Hydrogen Bonds
<b>Ethanol</b> $\text{CH}_3\text{CH}_2\text{OH}$ $12 + 5 + 16 + 1 + 1 = 30$	$\begin{array}{c} \text{H} \quad \text{H} \quad   \\   \quad   \quad   \\ \text{H} - \text{C} - \text{C} - \ddot{\text{O}} - \text{H} \\   \quad   \\ \text{H} \quad \text{H} \end{array}$	46.068	Hydrogen Bonds
<b>n-Butanol</b> $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$	$\begin{array}{c} \text{H} \quad \text{H} \quad \text{H} \quad \text{H} \quad \text{H} \\   \quad   \quad   \quad   \quad   \\ \text{H} - \text{C} - \text{C} - \text{C} - \text{C} - \ddot{\text{O}} - \text{H} \\   \quad   \quad   \quad   \\ \text{H} \quad \text{H} \quad \text{H} \quad \text{H} \end{array}$	74.122	Hydrogen Bonds
<b>Glycerin</b> $\text{C}_3\text{H}_8(\text{OH})_3$ $12 + 5 + 18 + 3 = 38$	$\begin{array}{c} \text{H} \quad \text{:}\ddot{\text{O}}\text{:} - \text{H} \quad \text{H} \\   \quad   \quad   \\ \text{H} - \text{C} - \text{C} - \text{C} - \text{H} \\   \quad   \quad   \\ \text{:}\ddot{\text{O}}\text{:} - \text{H} \quad \text{H} \quad \text{:}\ddot{\text{O}}\text{:} - \text{H} \end{array}$	92.093	Hydrogen Bonds
<b>Water</b> $\text{H}_2\text{O}$	$\begin{array}{c} \text{H} \\   \\ \text{H} - \ddot{\text{O}} - \text{H} \\   \\ \text{H} \end{array}$	18.015	Hydrogen Bonds

# Exploring Intermolecular Forces Lab Answers

**Norbert J. Pienta, Melanie M.  
Cooper, Thomas J. Greenbowe**



## Exploring Intermolecular Forces Lab Answers:

*Exploring General Chemistry in the Laboratory* Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science

*Exploring General, Organic, & Biochemistry in the Laboratory* William G. O'Neal, 2017-02-01 This full color comprehensive affordable manual is appropriate for two semester introductory chemistry courses It is loaded with clearly written exercises critical thinking questions and full color illustrations and photographs providing ample visual support for experiment set up technique and results

Exploring Chemistry (Loose-Leaf) Matthew Johll, 2012-05-04 Matthew Johll's *Exploring Chemistry* covers the standard topics for the nonmajors course in the typical order but each chapter unfolds in the context of a single case study that helps students connect what they are learning to real life situations For example students work through the often difficult topics of molecular structure gas laws and organic chemistry by learning about the development of powerful new chemotherapy drugs new technologies for screening airline passengers and the creation of biodegradable biopolymers It's the same case driven approach that Johll uses in his acclaimed *Investigating Chemistry* now in its Third Edition but *Exploring Chemistry* goes beyond the other book's specific focus on examples from forensic science to use real life stories from cooking athletics genetics green chemistry and more

**ChemDiscovery Lab Manual** Olga I. Agapova, 2002

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*Lab Adventures in Chemistry* Pasquale De Marco, 2025-03-09 Embark on an exciting journey into the world of chemistry with *Lab Adventures in Chemistry* a comprehensive guide that unravels the mysteries of matter and its transformations Discover the magic of chemistry as you explore the structure of atoms the properties of elements and compounds and the fascinating reactions that take place between them Through engaging experiments and real life examples this book brings chemistry to life making it accessible and enjoyable for students teachers and anyone curious about the world around them Delve into the wonders of chemical reactions explore the role of chemistry in everyday life and uncover the secrets of the chemical world With clear explanations and detailed illustrations *Lab Adventures in Chemistry* provides a solid foundation in the fundamentals of chemistry Learn about the periodic table chemical bonding energy changes and equilibrium and gain a deeper understanding of the properties and behavior of matter This book is more than just a textbook

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Blankenship, Rebecca J., Wiltsher, Cheree Y., Moton, Brandon A., 2022-06-24

Faculty learning communities are a fairly new ideology that is gaining traction among educators and institutions. These communities have numerous benefits on professional development, such as enhancing educator preparedness and learning. The possibilities of these communities are endless; however, further study is required to understand how these learning communities work and the best practices and challenges they face. *Experiences and Research on Enhanced Professional Development Through Faculty Learning Communities* shares the experiences and research related to the enhanced professional development received by university faculty and staff participating in a series of collaborative faculty learning communities. The book, using qualitative, quantitative, and mixed methodologies, considers educator experiences as participants in the faculty learning communities, what they learned, and how they applied and implemented best practices in their courses. Covering topics such as curricula, course design, and rubrics, this reference book is ideal for administrators, higher education professionals, program developers, program directors, researchers, academicians, scholars, practitioners, instructors, and students.

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