

### Activity A:

#### Subatomic particles

#### Get the Gizmo ready:

- Use the arrows to create an atom with two protons, two neutrons, and two electrons.
- Turn on **Show element name**.



**Question:** What are the properties of protons, neutrons, and electrons?

1. **Observe:** Turn on **Show element symbol** and **Element notation**. Three numbers surround the element symbol: the **mass number** ( $A$ ), electrical charge (no number is displayed if the atom is neutral), and the **atomic number** ( $Z$ ).
2. **Investigate:** Watch how the numbers change as you add or remove particles.



- A. Which number is equal to the number of protons in the atom?
- B. How can you calculate the number of neutrons ( $N$ ) in an atom?
- C. Which particle (proton, neutron, or electron) has a  
Positive charge?  
Negative charge?

neutrons

by the number of protons

electron

proton

neutron

3. **Explore:** Use the Gizmo to create different combinations of particles. Each combination represents a different element. Record the number of protons, neutrons, and electrons in each combination. The element is determined by the number of protons. The mass number is the sum of the number of protons and neutrons. The electrical charge is the difference between the number of protons and electrons.

- A. What are the mass numbers of hydrogen?
- B. What are the mass numbers of helium?
- C. What are the mass numbers of oxygen?

1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14, 15, 16, 17, 18, 19, 20, 21, 22, 23, 24, 25, 26, 27, 28, 29, 30, 31, 32, 33, 34, 35, 36, 37, 38, 39, 40, 41, 42, 43, 44, 45, 46, 47, 48, 49, 50, 51, 52, 53, 54, 55, 56, 57, 58, 59, 60, 61, 62, 63, 64, 65, 66, 67, 68, 69, 70, 71, 72, 73, 74, 75, 76, 77, 78, 79, 80, 81, 82, 83, 84, 85, 86, 87, 88, 89, 90, 91, 92, 93, 94, 95, 96, 97, 98, 99, 100

1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14, 15, 16, 17, 18, 19, 20, 21, 22, 23, 24, 25, 26, 27, 28, 29, 30, 31, 32, 33, 34, 35, 36, 37, 38, 39, 40, 41, 42, 43, 44, 45, 46, 47, 48, 49, 50, 51, 52, 53, 54, 55, 56, 57, 58, 59, 60, 61, 62, 63, 64, 65, 66, 67, 68, 69, 70, 71, 72, 73, 74, 75, 76, 77, 78, 79, 80, 81, 82, 83, 84, 85, 86, 87, 88, 89, 90, 91, 92, 93, 94, 95, 96, 97, 98, 99, 100

1

4. **Explore:** Use the Gizmo to answer the following questions.

- A. How many electrons are in a neutral atom of hydrogen?
- B. How many electrons are in a neutral atom of helium?
- C. What is the mass number of an atom with 3 protons and 3 neutrons?
- D. What is the mass number of an atom with 11 protons and 11 neutrons?
- E. How many electrons are in  $\text{He}^{2+}$ ?

1

2

6

22

11

# Explore Learning Element Builder Gizmo Answers Key

**Rachel Sandford**



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## **Table of Contents Explore Learning Element Builder Gizmo Answers Key**

1. Understanding the eBook Explore Learning Element Builder Gizmo Answers Key
  - The Rise of Digital Reading Explore Learning Element Builder Gizmo Answers Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Explore Learning Element Builder Gizmo Answers Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Explore Learning Element Builder Gizmo Answers Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Explore Learning Element Builder Gizmo Answers Key
  - Personalized Recommendations
  - Explore Learning Element Builder Gizmo Answers Key User Reviews and Ratings
  - Explore Learning Element Builder Gizmo Answers Key and Bestseller Lists
5. Accessing Explore Learning Element Builder Gizmo Answers Key Free and Paid eBooks
  - Explore Learning Element Builder Gizmo Answers Key Public Domain eBooks
  - Explore Learning Element Builder Gizmo Answers Key eBook Subscription Services
  - Explore Learning Element Builder Gizmo Answers Key Budget-Friendly Options

6. Navigating Explore Learning Element Builder Gizmo Answers Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Explore Learning Element Builder Gizmo Answers Key Compatibility with Devices
  - Explore Learning Element Builder Gizmo Answers Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Explore Learning Element Builder Gizmo Answers Key
  - Highlighting and Note-Taking Explore Learning Element Builder Gizmo Answers Key
  - Interactive Elements Explore Learning Element Builder Gizmo Answers Key
8. Staying Engaged with Explore Learning Element Builder Gizmo Answers Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Explore Learning Element Builder Gizmo Answers Key
9. Balancing eBooks and Physical Books Explore Learning Element Builder Gizmo Answers Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Explore Learning Element Builder Gizmo Answers Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Explore Learning Element Builder Gizmo Answers Key
  - Setting Reading Goals Explore Learning Element Builder Gizmo Answers Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Explore Learning Element Builder Gizmo Answers Key
  - Fact-Checking eBook Content of Explore Learning Element Builder Gizmo Answers Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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