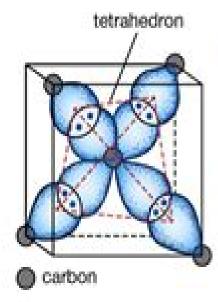
Chemical bonding in crystalline solids

+ + +

ionic bond

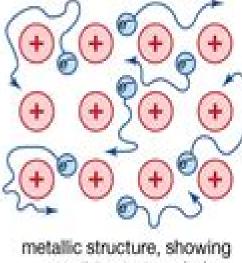
An idealized ionic (or electrovalent) bonding of oppositely charged ions.

covalent bond



Covalent bonds involve electron sharing, such as between these carbon atoms when they form a diamond,

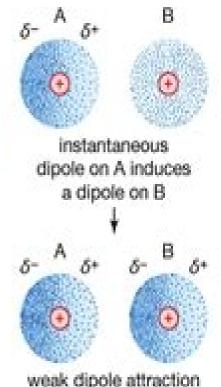
metallic bond



metallic structure, showing possible electron (e⁻) paths around the nuclei of metal atoms (represented as spheres with a positive charge)

Metallic bonding can be thought of as a cloud of positively charged ions immersed in a cloud of valence electrons.

van der Waals bond



Neutral molecules may be held together by a weak electric force known as the van der Waals bond.

of van der Waals bond

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