

Naming Chemical Compounds Worksheet

Name the following *ionic* compounds:

- 1) NaBr Sodium Bromide
- 2) CaO Calcium Oxide
- 3) Li_2S Lithium Sulfide
- 4) MgBr_2 _____
- 5) $\text{Pu}(\text{OH})_3$ _____
- 6) $\text{Hg}(\text{CN})_2$ _____
- 7) $\text{Mo}(\text{C}_2\text{H}_3\text{O}_2)$ _____
- 8) $\text{Cr}(\text{CrO}_4)_3$ _____
- 9) WO_2 _____
- 10) $\text{Mg}(\text{ClO})_2$ _____

Write the formulas for the following *ionic* compounds:

11. potassium dichromate _____
12. gold(III) oxide _____
13. aluminum periodate _____
14. barium nitrite _____
15. silver carbonate _____
16. lithium sulfite _____
17. zinc hydrogen carbonate _____
18. iron(III) hydroxide _____
19. ammonium phosphate _____
20. copper(II) bromite _____
- 21) SO_3 _____
- 22) N_2S _____
- 23) BF_3 _____
- 24) P_2Br_4 _____
- 25) SiO_2 _____
- 26) SF_6 _____
- 27) NO_2 _____
- 28) nitrogen trichloride _____
- 29) dinitrogen trioxide _____
- 30) phosphorus pentafluoride _____
- 31) diboron tetrahydride _____

Holt Unit 7 Bonding Naming And Chemical Reactions Test

E Durkheim



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Table of Contents Holt Unit 7 Bonding Naming And Chemical Reactions Test

1. Understanding the eBook Holt Unit 7 Bonding Naming And Chemical Reactions Test
 - The Rise of Digital Reading Holt Unit 7 Bonding Naming And Chemical Reactions Test
 - Advantages of eBooks Over Traditional Books
2. Identifying Holt Unit 7 Bonding Naming And Chemical Reactions Test
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Holt Unit 7 Bonding Naming And Chemical Reactions Test
 - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Unit 7 Bonding Naming And Chemical Reactions Test
 - Personalized Recommendations
 - Holt Unit 7 Bonding Naming And Chemical Reactions Test User Reviews and Ratings
 - Holt Unit 7 Bonding Naming And Chemical Reactions Test and Bestseller Lists
5. Accessing Holt Unit 7 Bonding Naming And Chemical Reactions Test Free and Paid eBooks
 - Holt Unit 7 Bonding Naming And Chemical Reactions Test Public Domain eBooks
 - Holt Unit 7 Bonding Naming And Chemical Reactions Test eBook Subscription Services
 - Holt Unit 7 Bonding Naming And Chemical Reactions Test Budget-Friendly Options

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 - ePub, PDF, MOBI, and More
 - Holt Unit 7 Bonding Naming And Chemical Reactions Test Compatibility with Devices
 - Holt Unit 7 Bonding Naming And Chemical Reactions Test Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Holt Unit 7 Bonding Naming And Chemical Reactions Test
 - Highlighting and Note-Taking Holt Unit 7 Bonding Naming And Chemical Reactions Test
 - Interactive Elements Holt Unit 7 Bonding Naming And Chemical Reactions Test
8. Staying Engaged with Holt Unit 7 Bonding Naming And Chemical Reactions Test
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Holt Unit 7 Bonding Naming And Chemical Reactions Test
9. Balancing eBooks and Physical Books Holt Unit 7 Bonding Naming And Chemical Reactions Test
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Holt Unit 7 Bonding Naming And Chemical Reactions Test
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Holt Unit 7 Bonding Naming And Chemical Reactions Test
 - Setting Reading Goals Holt Unit 7 Bonding Naming And Chemical Reactions Test
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Holt Unit 7 Bonding Naming And Chemical Reactions Test
 - Fact-Checking eBook Content of Holt Unit 7 Bonding Naming And Chemical Reactions Test
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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