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#### Balancing Chemical Equations – Homework Sheet Grade 10 Science

# PART 1: Balance the following chemical equations \*Note, you may need to work out these balancing equations on extra paper

N<sub>2</sub> + H<sub>2</sub> → NH<sub>3</sub>

5<sub>8</sub> + O<sub>2</sub> → 5O<sub>3</sub>

3.  $HgO \rightarrow Hg + O_2$ 

Zn + HCl → ZnCl<sub>2</sub> + H<sub>2</sub>

5. SiCl<sub>4</sub> + H<sub>2</sub>O → H<sub>4</sub>SiO<sub>4</sub> + HCl

6. Na +  $H_2O \rightarrow NaOH + H_2$ 

H<sub>3</sub>PO<sub>4</sub> → H<sub>4</sub>P<sub>2</sub>O<sub>7</sub> + H<sub>2</sub>O

Si<sub>2</sub>H<sub>3</sub> + O<sub>2</sub> → SiO<sub>2</sub> + H<sub>2</sub>O

Al(OH)<sub>3</sub> + H<sub>2</sub>SO<sub>4</sub> → Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> + H<sub>2</sub>O

Fe + O<sub>2</sub> → Fe<sub>2</sub>O<sub>3</sub>

Fe<sub>2</sub>(5O<sub>4</sub>)<sub>3</sub> + KOH → K<sub>2</sub>SO<sub>4</sub> + Fe(OH)<sub>3</sub>

FeS<sub>2</sub> + O<sub>2</sub> → Fe<sub>2</sub>O<sub>3</sub> + SO<sub>2</sub>

Al + FeO → Al<sub>2</sub>O<sub>3</sub> + Fe

Na<sub>2</sub>CO<sub>3</sub> + HCI → NaCI + H<sub>2</sub>O + CO<sub>2</sub>

15. K + Br<sub>2</sub> → KBr

16. P<sub>4</sub> + O<sub>2</sub> → P<sub>2</sub>O<sub>5</sub>

17. C<sub>2</sub>H<sub>2</sub> + O<sub>2</sub> → CO<sub>2</sub> + H<sub>2</sub>O

# **Grade 9 Natural Science Balancing Equations**

**Richard Bailey** 

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