Stoichiometry (stoich)

12/04

Stoichiometry describes the method of measuring the elements of a chemical reaction. This method allows you to determine how much of all the other chemicals you will need or can produce when you know anything about the amount of at least one chemical in the reaction. You will definitely need a correct balanced equation in order to use this method, since it requires the use of mole ratios.

EXAMPLE:

If you burn 5.5 mol of Catter, now many mol of oxygen will you need to an this?

* Remember to use correct # of sig figs when reporting your answer! 19 mol 02

* cancel units to ensure your answer has

Lecture Notes On Stoichiometry

Roman Wölfel

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