Chest Sheet

- molar mass is for a compound (g/mol), e.g. CO₂ = 44.01 g/mol
- atomic mass is for an element (pinol), e.g Al = 26.98 g/mol

Scenario 1: calculating moles from grams

grams of X
$$\frac{1 \text{ mol of X}}{\text{grams of X}}$$
 (divide by the atomic mass); the atomic mass isn't affected by the amount of moles in the chemical equation

88.02 g of $C\Theta_2$ = 2 moles of CO_2

44.01 $\text{pol}(CO_3)$

Scenario 2: calculating grams from moles

Learning Guide To Stoichiometry

Marcel A. Müller

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