

Cellular Respiration

SHORT ANSWER

Formulation of CO₂ from Sugars by Yeast Cells

Lab Results:

- The graph of the sugar fermentation by yeast. All listed other factors for determine the amount of CO₂ produced.

Sugars Type	Substrate Amount and structure (mol)	Product Yielded at 15 minutes (mol)	Amount of CO ₂ produced at that condition
glucose	0.001	0.01	0.01
fructose	0.001	0.01	0.01
sucrose	0.001	0.01	0.01
maltose	0.001	0.01	0.01

- The graph of the sugar fermentation by yeast. All listed other factors for determine the rate of sugar fermentation per volume of starting fermentation. They should have 1 for substrate for glucose from the multiple choice word bank.
 - To calculate the value of CO₂ produced per (for substrate) available, assuming P = 1 atm.
 - at 15 min
 - value P is the atmosphere is pressure at the lab. It is the volume in liter of CO₂ mole produced at volume of CO₂ at the time gas collected (L-CO₂ is, atmosphere) and P is the temperature in Kelvin.
 - To calculate the value of sugar converted use the following equation:

$$n = PV/RT$$
 - value n is the amount of moles of sugar converted, or in this number of moles of CO₂ produced, and this the amount of sugar in that sugar.
 - To calculate the rate of sugar fermentation per volume use the following equation:

$$r = (dCO_2/dt)/V$$
 - value r is the rate of sugar fermentation per volume, or is the amount of moles of sugar converted, or is the mole fraction weight of the sugar in system, which is the composition being investigated.

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WJ Hussar



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