

## CHEM 101 Fall 10 Exam I(a)

On the answer sheet (scantron) write your name, student ID number, and recitation section number. Choose the best (most correct) answer for each question and enter it on your answer sheet.

Potentially useful data:

Avogadro's constant:  $6.022 \times 10^{23}$

Speed of light:  $c = 3.00 \times 10^8$  m/s

Planck constant:  $h = 6.626 \times 10^{-34}$  J s

$E = hc/\lambda$

1. Please choose the letter "a" as your answer for this question.
2. Choose the **incorrect** statement: An explanation of a phenomenon, in the form of a properly formulated scientific hypothesis,
  - a. should be cleverly constructed so as to resist all future attempts to falsify it.
  - b. should have predictive capability.
  - c. should be rejected or changed if future experimental results disagree with it.
  - d. may turn out to be disproved by subsequent experiments.
  - e. may not be immediately accepted by all scientists in the field.
3. Which characteristic below best fits the description of a solid?
  - a. large distances between the molecules
  - b. molecules that are close together but are moving past one another
  - c. highly disordered molecules
  - d. rapid molecular motion
  - e. highly ordered molecules
4. Which of the following is correct?
  - a.  $^{81}\text{Br}$  has 35 protons, 35 electrons and 81 neutrons.
  - b.  $^{34}\text{S}$  has 16 protons, 16 electrons and 16 neutrons.
  - c.  $^{40}\text{Ar}$  has 18 protons, 18 electrons and 12 neutrons.
  - d.  $^{59}\text{Co}$  has 27 protons, 27 electrons and 32 neutrons.
  - e.  $^{48}\text{Ti}$  has 22 protons, 22 electrons and 24 neutrons.
5. Which of the following represent a pair of isotopes?

|                       |                       |                     |                     |                       |
|-----------------------|-----------------------|---------------------|---------------------|-----------------------|
| $^{37}_{17}\text{Cl}$ | $^{37}_{18}\text{Ar}$ | $^{17}_{8}\text{O}$ | $^{17}_{9}\text{F}$ | $^{40}_{18}\text{Ar}$ |
| I                     | II                    | III                 | IV                  | V                     |

  - a. I and II
  - b. I and III
  - c. I and IV
  - d. II and V
  - e. III and IV

# Gen Chemistry Semester Review Answers

**SA Adler**



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