Name: _____ Class: _____ Period: _____

Partnership #11 – The Periodic and the Submicroscopic (Electrons in Rings)

Scientists believe that we need to talk about the (SAME) old things, but in a new way. We need to stop thinking about the submicroscopic world as a place where electrons orbit nuclei in circular paths, and start thinking about it as a place where electrons behave in unpredictable ways.

Part 1: Electron Density **Imagine that you have a model of an electron in a ring. Describe what you would see if you looked at the ring from the side.**	
Part 2: Electron Density **Imagine that you have a model of an electron in a ring. Describe what you would see if you looked at the ring from the top.**	
Part 3: Electron Density **Imagine that you have a model of an electron in a ring. Describe what you would see if you looked at the ring from the side and from the top.**	
Part 4: Electron Density **Imagine that you have a model of an electron in a ring. Describe what you would see if you looked at the ring from the side and from the top.**	
Part 5: Electron Density **Imagine that you have a model of an electron in a ring. Describe what you would see if you looked at the ring from the side and from the top.**	
Part 6: Electron Density **Imagine that you have a model of an electron in a ring. Describe what you would see if you looked at the ring from the side and from the top.**	
Part 7: Electron Density **Imagine that you have a model of an electron in a ring. Describe what you would see if you looked at the ring from the side and from the top.**	
Part 8: Electron Density **Imagine that you have a model of an electron in a ring. Describe what you would see if you looked at the ring from the side and from the top.**	

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