

Energy Changes in Chemical reactions for KSS Science - Worksheet (Answer)

- In an exothermic reaction does the temperature go up or down?
The temperature will increase
 - In an endothermic reaction does the temperature go up or down?
The temperature will decrease
 - Name two examples of exothermic reactions.
Respiration & combustion
 - Name two examples of endothermic reactions.
Photosynthesis & thermal decomposition
 - Copy the correct answers.
The words between the arrows of the reactions usually mean to the student that the
is an exothermic reaction. Therefore, the words used inside between the arrows
of the reactions usually, are in an endothermic reaction.
 - Use the table to answer the question.
- | Reaction | Starting temperature /°C | Final temperature /°C |
|----------|--------------------------|-----------------------|
| (a) | 20 | 30 |
| (b) | 20 | 10 |
| (c) | 20 | 20 |
- Which reaction was exothermic? endothermic? exothermic, reaction heat was
released.
 a. Exothermic temperature increases
 b. Endothermic temperature decreases
 c. Exothermic temperature increases
 - Which reaction has the biggest energy change?
 a. biggest temperature change
 - In an exothermic reaction, is enthalpy change positive or negative?
negative
 - In an endothermic reaction, is enthalpy change positive or negative?
positive
 - When hydrochloric acid reacts with ammonium hydroxide to produce the
 reaction: $\text{HCl} + \text{NH}_4\text{OH} \rightarrow \text{NH}_4\text{Cl} + \text{H}_2\text{O}$
 given: $\Delta H = -50 \text{ kJ/mol}$
 Complete the energy profile diagram and state whether the reaction is exothermic
 or endothermic, explain your answer.

Energy And Chemical Change Answers

Clemens Wendtner

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Energy And Chemical Change Answers:

Energy And Chemical Change Answers Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Energy And Chemical Change Answers**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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