

Empirical and Molecular Formula Practice KEY

Empirical Formula

1. 36.5% Na, 25.4% S, and 38.1% O forms this compound.

$$\text{Na} - 36.5/23 = 1.56/0.8 = 2$$

$$\text{S} - 25.4/32 = 0.8/0.8 = 1$$

$$\text{O} - 38.1/16 = 2.4/0.8 = 3$$



2. A compound made of 72.2% Magnesium and 27.8% nitrogen by mass.

$$\text{Mg} - 72.2/24.3 = 3/2 = 1.5(2) = 3$$

$$\text{N} - 27.8/14 = 2/2 = 1(2) = 2$$



3. A chromium compound with 68.4% Cr and 31.6% O.

$$\text{Cr} - 68.4/52 = 1.315/1.315 = 1(2) = 2$$

$$\text{O} - 31.6/16 = 1.975/1.315 = 1.5(2) = 3$$



4. A compound with the composition of 24% C, 7% H, 38% F, and 31% P.

$$\text{C} - 24/12 = 2/1 = 2$$

$$\text{H} - 7/1 = 7/1 = 7$$

$$\text{F} - 38/19 = 2/1 = 2$$

$$\text{P} - 31/31 = 1/1 = 1$$



5. 63.5% silver, 8.2% nitrogen, and 28.3% oxygen make up this compound.

$$\text{Ag} - 63.5/107.87 = 0.589/0.585 = 1$$

$$\text{N} - 8.2/14.01 = 0.585/0.585 = 1$$

$$\text{O} - 28.3/16 = 1.769/0.585 = 3$$



6. 8.81g C and 91.2g Chlorine forms this compound.

$$\text{C} - 8.81/12 = 0.73/0.73 = 1(2) = 2$$

$$\text{Cl} - 91.2/35 = 2.61/0.73 = 3.5(2) = 7$$



7. A compound in a soft drink that has 0.31g hydrogen, 3.2g phosphorus, and 6.5g oxygen.

$$\text{H} - 0.31/1 = 0.31/.1 = 3.1$$

$$\text{P} - 3.2/31 = 0.10/.1 = 1$$

$$\text{O} - 6.5/16 = 0.40/.1 = 4$$



Empirical And Molecular Formula Key

**Morris Hein, Scott Pattison, Susan
Arena, Leo R. Best**



Empirical And Molecular Formula Key:

PHARMACEUTICAL ORGANIC CHEMISTRY-I SHAIK MUNWAR, SHAIK KHADAR YAZDAN, SRIDEVI

GUDIVADA,2024-09-25 PREFACE Pharmaceutical Organic Chemistry is a vital branch of organic chemistry that focuses on the preparation structure and reactions of organic compounds with particular emphasis on their application in pharmaceuticals This field is crucial because it encompasses all chemical reactions related to life processes making its study essential for understanding and developing new pharmaceutical substances The evolution of Pharmaceutical Organic Chemistry stems from its application in drug development integrating knowledge from organic chemistry into practical uses for pharmaceuticals Organic chemistry provides the foundation for biochemistry which explores health and disease and is critical for the practice of nutritional medical and related life sciences It also underpins advancements in medicinal chemistry bioinformatics biotechnology gene therapy pharmacology pathology chemical engineering dental science and more Understanding organic chemistry helps in identifying the reactivity of compounds predicting their reactions and designing substances with desired properties This knowledge is instrumental in various careers including those of doctors engineers pharmacists veterinarians dentists pharmacologists and chemists Thus a solid grasp of organic chemistry is essential for success in these fields Despite its importance organic chemistry is often perceived as challenging This perception raises questions such as How should one start learning organic chemistry What should be studied and How can one effectively remember chemical reactions This book aims to address these concerns by offering a comprehensive guide that simplifies the study of Pharmaceutical Organic Chemistry Instead of rote memorization this book encourages understanding the subject conceptually It is designed to make learning organic chemistry engaging and enjoyable **Introduction to General,**

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