

Empirical and Molecular Formula Practice

3 pages + 3-page Answer KEY + Periodic Table

EMPIRICAL AND MOLECULAR PROBLEMS

Chemistry

1 The percent composition of a certain gas is nitrogen (N) 82.1% and hydrogen (H) 17.9%. Calculate the empirical formula.
Show & Convert is to %

Step 2: Convert % to mol

Step 3: Divide by the smallest mole value

Step 4: Multiply by smallest number that will make a whole

2 The compound in problem 1 has a molar mass of 80 g/mol. Calculate the molecular formula.

Step 5: Divide molecular mass (mol) by empirical mass

Step 6: Multiply empirical formula by the answer in Step 5

3 A compound is composed of 60.1% nitrogen and 39.9% hydrogen. What is the empirical formula of this compound?

4 The compound in problem 3 has a molar mass of 100 g/mol. Calculate the molecular formula.

5 Hydrogen, a high-energy fuel, is 14.1% hydrogen. Calculate the empirical formula.

6 Hydrogen's molar mass is 100 g/mol. Calculate the molecular formula.

7 A compound composed of hydrogen (1.0%) and oxygen (99.0%) has a molar mass of 34 g/mol. What is the empirical formula of this compound?

8 What is the molecular formula of the compound in problem 7?

9 A compound with a known molar mass of 100 g/mol contains carbon (60.1%) and hydrogen (39.9%). What is the empirical formula?

10 Calculate the molecular formula of the compound in problem 9.

ANSWER KEY
INCLUDED

Empirical And Molecular Calculation Practice Packet

Answer Key

ML Morrison



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Reviewing **Empirical And Molecular Calculation Practice Packet Answer Key**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Empirical And Molecular Calculation Practice Packet Answer Key**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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