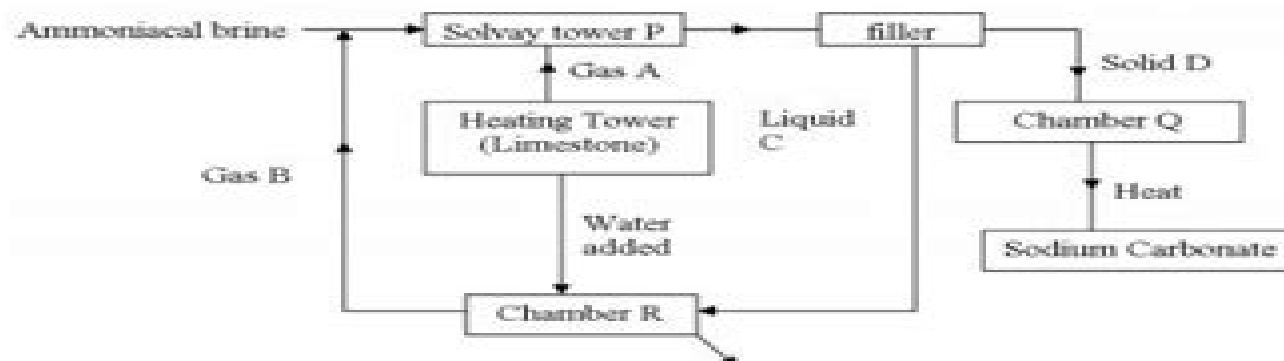


2. The scheme below shows the manufacture of sodium carbonate by the Solvay process. Study it and use it to answer the questions that follow.



(a) Name (i) gases A and B (1mark)

(b) Name liquid C and Solid D (1mark)

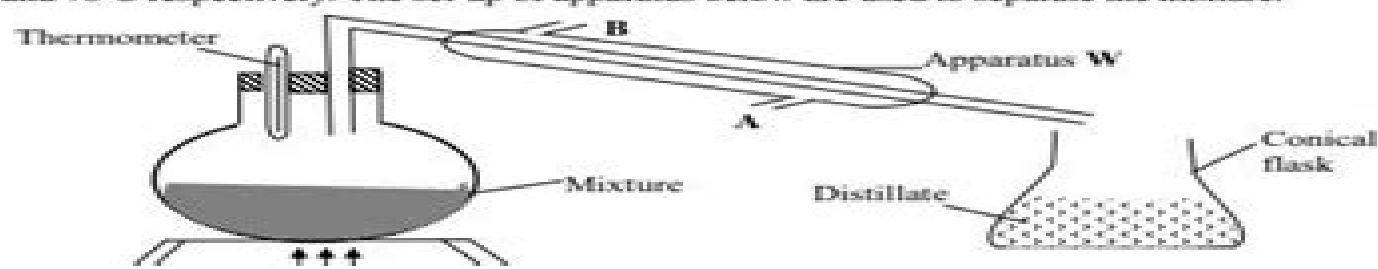
(c) Write equations for the reactions taking place in tower P and chamber R (2marks)

(d) Name the product formed in chamber at chamber R and give one of its uses (2marks)

.....

(c) State two uses of sodium carbonate (1mark)

3. A student left some crushed fruit mixture with water for some days. He found the mixture had fermented. He concluded that the mixture was contaminated with water and ethanol with boiling point of 100°C and 78°C respectively. The set-up of apparatus below are used to separate the mixture.



High School Chemistry Form 3

Y Pai



High School Chemistry Form 3:

This book delves into High School Chemistry Form 3. High School Chemistry Form 3 is a crucial topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into High School Chemistry Form 3, encompassing both the fundamentals and more intricate discussions.

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- Chapter 3: High School Chemistry Form 3 in Everyday Life
- Chapter 4: High School Chemistry Form 3 in Specific Contexts
- Chapter 5: Conclusion

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