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What Now
with Nicole D'Dell



Nicole D'Dell

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web answer a 1 explanation the entire reaction was multiplied by 2 therefore Δh must also be halved 3 answer c explanation option c is incorrect because 566 kJ/mol is the Δh per mole of O_2 and not for 2 moles even though options b and d are expressed in grams they are still correct because 56

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 $2C(s) + 3O_2(g) \rightarrow 2CO_2(g)$ $\Delta H = -7914 \text{ kJ}$ is this endothermic or exothermic?

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