

Polyatomic Ions Worksheet

Polyatomic Ions

I. Name or write the formula for the following polyatomic ions:

sulfate	_____	CO_3^{2-}	_____
nitrite	_____	SO_4^{2-}	_____
perchlorate	_____	BrO_3^-	_____
hypochlorite	_____	CO_3^{2-}	_____
chlorate	_____	PO_4^{3-}	_____
phosphate	_____	ClO^-	_____
percarbonate	_____	NO_2^-	_____
chromate	_____	PO_4^{3-}	_____
hyposulfite	_____	HCO_3^-	_____
permanganate	_____	SO_4^{2-}	_____
carbonate	_____		

II. Name or write the formula for the following "Type I" polyatomic ionic compounds:

barium hydrosulfide	_____	$\text{Ba}(\text{OH})_2$	_____
sodium nitrite	_____	$\text{Ca}(\text{OH})_2$	_____
ammonium chloride	_____	Ag_2SO_4	_____
calcium bisulfate	_____	MgCO_3	_____
rubidium perchlorate	_____	$\text{Sr}_2\text{H}_2\text{O}_4$	_____
chromium sulfate	_____	$\text{Al}(\text{C}_2\text{H}_3\text{O}_2)_3$	_____
aluminum acetate	_____	SrSO_4	_____
ammonium nitrate	_____	$\text{Fe}(\text{OH})_2$	_____
magnesium hydrogencarbonate	_____	$\text{Ca}(\text{HCO}_3)_2$	_____
silver hypochlorite	_____	Sr_2CO_3	_____
potassium cyanide	_____	NaNO_2	_____
barium iodate	_____	$\text{Ba}(\text{OH})_2$	_____

Lots Of Ionic Naming Practice Problems Key

JL Elias



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