

## Journey to Topaz by Yoshiko Uchida

### Reading Comprehension Questions

1. Who is the author of "Journey to Topaz"?
2. In which historical period is the story set?
3. Where does the main character, Yuki, live at the beginning of the book?
4. What event triggers Yuki's family's relocation?
5. What is the name of the internment camp where Yuki and her family are sent?
6. Who is Yuki's best friend in her old neighborhood?
7. What is the main reason given for the internment of Japanese-Americans during World War II?
8. How does Yuki feel about leaving her home?
9. What possessions are Yuki and her family forced to leave behind?
10. How do the characters travel to the internment camp?
11. What is Yuki's younger brother's name?
12. How do the characters adapt to their new living conditions in the camp?
13. What kind of work does Yuki's father do in the camp?
14. Who is Mr. Jamison, and what role does he play in the camp?
15. How does Yuki feel about her new school in the internment camp?
16. What is the significance of the Marianne Free Press?
17. What role does art play in Yuki's life and in the camp?
18. How do Yuki and her family cope with the harsh living conditions in the camp?
19. Who is Ken, and what is his relationship with Yuki?
20. Describe Yuki's relationship with her grandmother.
21. What does Yuki's family do to celebrate Christmas in the internment camp?
22. How do the characters celebrate the New Year?
23. Who are the antagonists in the story?
24. What are the main challenges Yuki faces while living in the internment camp?
25. How does Yuki feel about her Japanese heritage?
26. What is the significance of Yuki's diary?
27. How does Yuki's perception of America change throughout the story?
28. What happens to Yuki's father at the end of the book?
29. How does Yuki's family finally leave the internment camp?
30. What are Yuki's dreams for the future?
31. What role does the theme of identity play in the story?

# Journey To Topaz Comprehension Answer

**H Kauffman**



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web oct 6 2022 therefore boyle s law indicates that when the pressure of the object increases the volume of the object decreases and vice versa because volume and pressure s relationship is inversely proportional while the temperature is constant

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web we then changed the volume between 5ml and 20ml and recorded the pressure at each this part of the lab showed that volume and pressure are inversely proportionate as the pressure would decrease if the volume was increased part b of the experiment looked at the relationship between temperature and pressure

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web procedure 1 the relationship between pressure and volume set the plunger of the syringe to 1 10 the capacity of the syringe the 2 0 ml mark if using a 20 ml syringe the volume is read from the edge of the bottom rib of the plunger attach the syringe to the connector at the end of sensor

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414 l 22 414 l we can substitute 101 325 kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web may 20 2018 french physicist jacques charles 1746 1823 studied the effect of temperature on the volume of a gas at constant pressure charles s law states that the volume of a given mass of gas varies directly with the absolute temperature of the gas when pressure is kept constant the absolute temperature is temperature measured

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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web figure pageindex 10 since the number of moles in a given volume of gas varies with pressure and temperature changes chemists use standard temperature and pressure 273 15 k and 1 atm or 101 325 kpa to report properties of gases

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introduction the ideal gas law is a commonly used formula in chemistry it relates pressure temperature and volume this  
experiment seeks to prove those relationships by measuring the pressure of gas at various temperatures

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web the pressure volume temperature pvt fluids lab in ku s chemical petroleum engineering department is used to study the  
properties of fluids under a wide range of pressure and temperature

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web volume pressure kpa linear pressure kpa volume ml pressure kpa in part b we observed the relationship between  
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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of  
declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95  
82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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surface and interfacial tension ph refractive index and vapor pressure is performed in the pvt laboratory pressure volume and  
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