

Heat Transfer

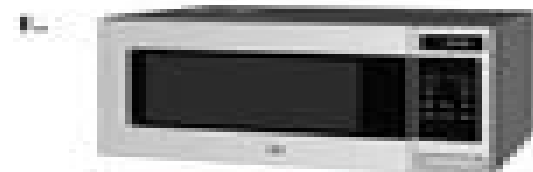
Using words from the word boxes below, complete the paragraph about heat transfer.

fastest	hot	low	solid	fluid
conduction	above	convection	energy	emit
warmer	matter	transfer	absorb	temperature
radiation	contact	cold	vibrate	waves

All matter has heat. Heat is a form of energy caused by particles in an object that vibrate. The faster the particles of an object vibrate, the warmer the object will be. Because particles of an object are always moving, heat transfer is always happening. Heat always flows in the same direction: from hot to cold. Heat transfer will stop once two objects reach the same temperature. This is known as equilibrium.

There are three key ways that heat transfers. With solid objects, heat transfers when the objects come into direct contact with other things. This is known as conduction. Liquids and gases are different. Because these two states of matter flow, or are fluid, heat transfer happens when warmer, less dense particles rise and cooler, more dense particles sink. This ongoing process is known as a convection current. Heat can also be transferred through space (distance) in the form of waves. This process is known as radiation. All objects give off, or emit, some heat. All objects also take in, or absorb, heat.

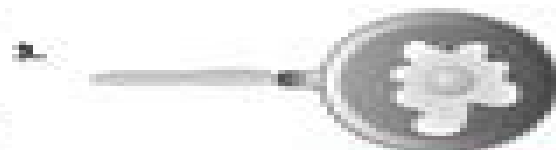
Identify the method of heat transfer that takes place in each illustration. Write the method of heat transfer underneath the picture.



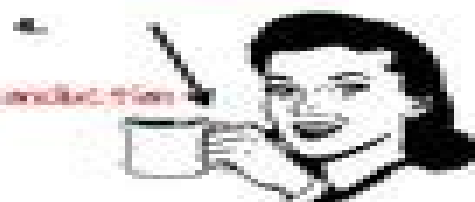
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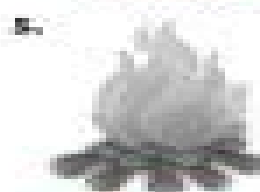
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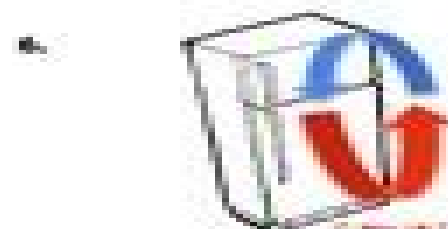
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of the text examines ABET Accreditation Board for Engineering and Technology related topics of concern including economies and finance numerical methods open ended problems ethics environmental management and safety and accident management

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Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many

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