

1. What amount of oxygen,  $O_2$ , (in moles) contains  $1.8 \times 10^{22}$  molecules?

- |                         |       |                         |        |
|-------------------------|-------|-------------------------|--------|
| <input type="radio"/> A | 0.030 | <input type="radio"/> B | 3.0    |
| <input type="radio"/> C | 0.30  | <input type="radio"/> D | 0.0030 |

2. Which compound has the empirical formula with the greatest mass?

- |                         |             |                         |             |
|-------------------------|-------------|-------------------------|-------------|
| <input type="radio"/> A | $C_6H_6$    | <input type="radio"/> B | $C_5H_{10}$ |
| <input type="radio"/> C | $C_4H_{10}$ | <input type="radio"/> D | $C_2H_6$    |

3.  $\_\_C2H2(g) + \_\_O2(g) \rightarrow \_\_CO2(g) + \_\_H2O(g)$

When the equation above is balanced, what is the coefficient for oxygen?

- |                         |   |                         |   |
|-------------------------|---|-------------------------|---|
| <input type="radio"/> A | 2 | <input type="radio"/> B | 3 |
| <input type="radio"/> C | 5 | <input type="radio"/> D | 4 |

4. 3.0 dm<sup>3</sup> of sulfur dioxide is reacted with 2.0 dm<sup>3</sup> of oxygen according to the equation below.



What volume of sulfur trioxide (in dm<sup>3</sup>) is formed? (Assume the reaction goes to completion and all gases are measured at the same temperature and pressure.)

- |                         |     |                         |     |
|-------------------------|-----|-------------------------|-----|
| <input type="radio"/> A | 5.0 | <input type="radio"/> B | 3.0 |
| <input type="radio"/> C | 2.0 | <input type="radio"/> D | 4.0 |

5. What will happen to the volume of a fixed mass of gas when its pressure and temperature (in Kelvin) are both doubled?

- |                         |                     |                         |                                 |
|-------------------------|---------------------|-------------------------|---------------------------------|
| <input type="radio"/> A | It will not change. | <input type="radio"/> B | The change cannot be predicted. |
| <input type="radio"/> C | It will increase.   | <input type="radio"/> D | It will decrease.               |

6. What amount (in moles) is present in 2.0 g of sodium hydroxide, NaOH?

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**Arshad Iqbal**



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